

## 5) "MAKE FIVE" A game about mineral recipes

This game is recommended for older students, or those who are very enthusiastic about rocks and minerals. If "Symbol Jars" was enough, you can skip this game. You could also wait and play this game after the next chapter.

By definition, a mineral has a definite chemical composition (a recipe). In this game you will be introduced to the recipes for some common minerals. It's also an opportunity to keep on learning all those letter abbreviations (symbols).

You will need: copies of the pattern pages copied onto card stock, scissors, and white glue (if you are assembling the paper dice) If you are using wooden cubes for the dice, you'll also need one or more markers.  
(In a pinch for time, just take a fine point marker (red?) and write on real dice. Everyone can ignore the dots.)

NOTE: If you can get three wooden cubes, this is the best option. Most craft stores sell wooden cubes by the "each" or in small units and fairly inexpensively. If you want this game sturdy enough to survive future uses, consider using wooden cubes.

### Preparation:

1) Cut out the dice patterns (copied onto heavy card stock) and make into cubes, using small dabs of white glue on the tabs.  
(Or, write the symbols on wooden dice or even regular dice.)

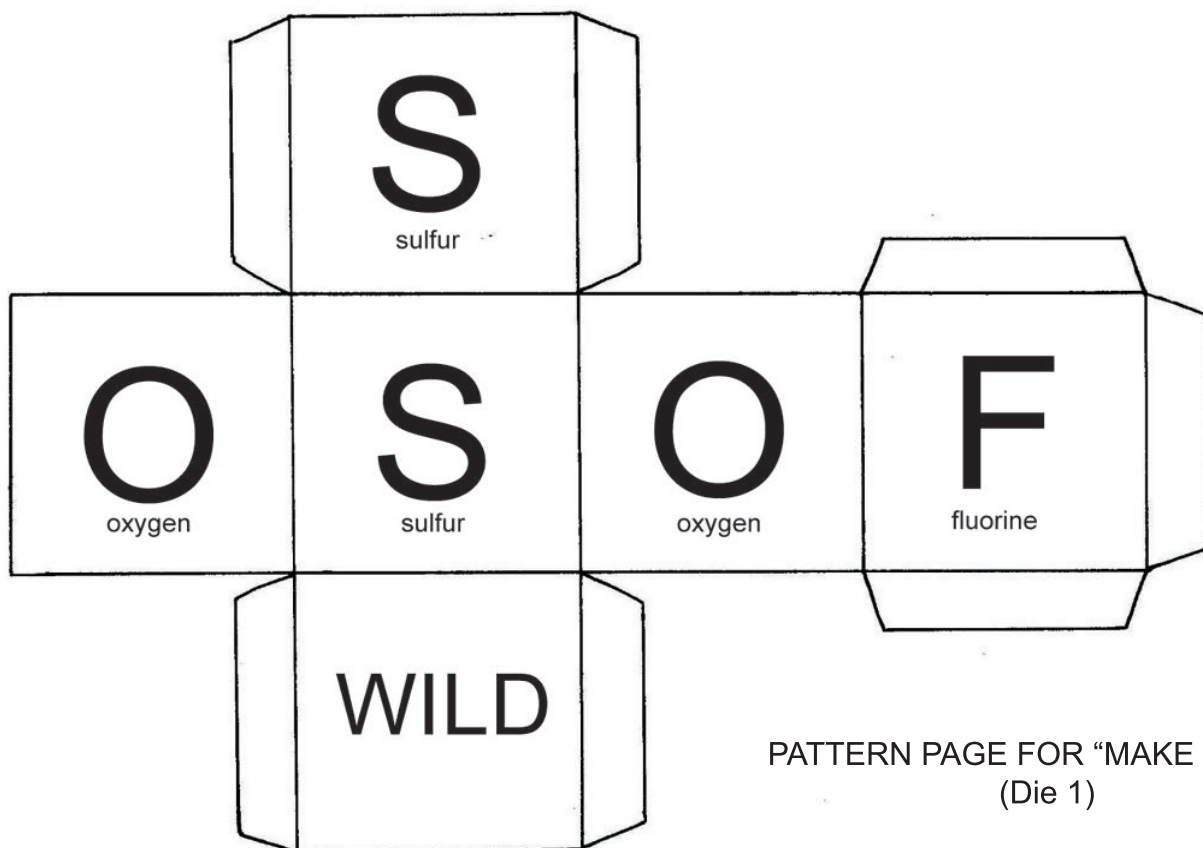
2) Cut apart the 16 mineral cards.

### How to play:

Place the mineral cards on the table, face up, so they form a 4 x 4 square. Each player will have a turn rolling all three dice at once. The goal is to roll the ingredients to form a mineral. (One roll of the three dice per player per turn.) For example, if the first player rolls: Cu, Fe, and S, he should notice that those are the ingredients of chalcopyrite. Therefore, that player picks up the chalcopyrite card. If the next player rolls Ca, C, and WILD, he could make the wild card into O, and be eligible to pick up calcite.

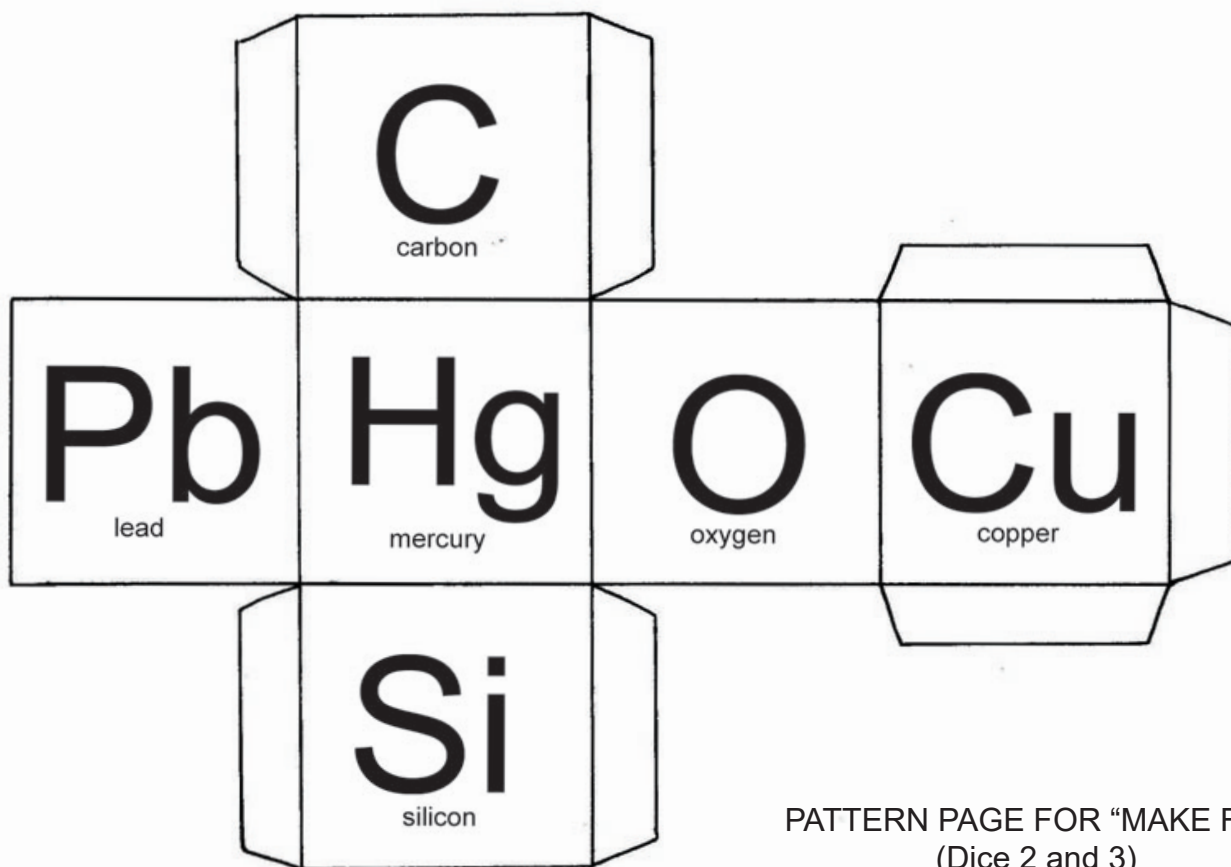
The first player to collect five cards wins the game.

NOTE: If you are working from a paperback copy of this book, not a digital download, and you would like a digital file so that you can print these patterns using your computer's printer, go to [www.ellenjmchenry.com](http://www.ellenjmchenry.com), click on FREE DOWNLOADS, then on CHEMISTRY, and then you will see a link for "Printable pages for The Elements curriculum."



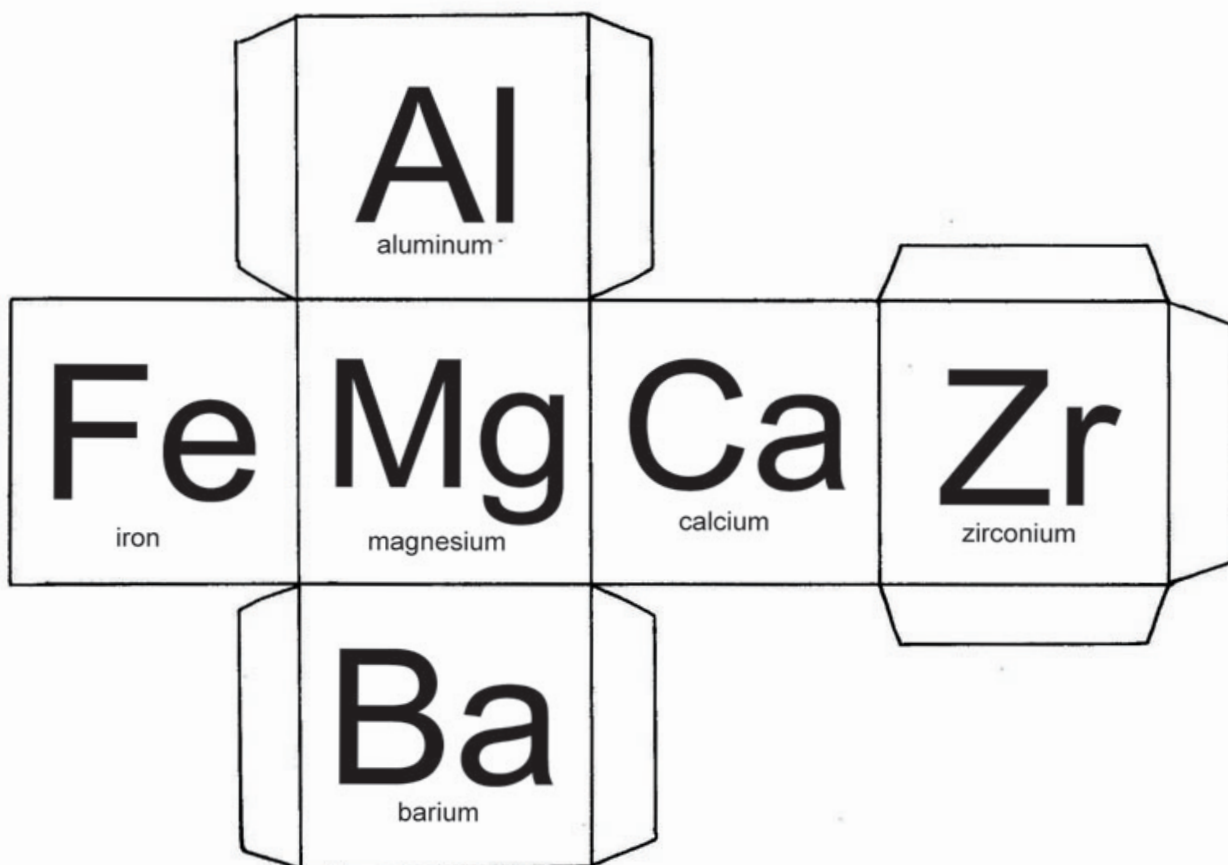
PATTERN PAGE FOR "MAKE FIVE"  
(Die 1)

COPY ONTO CARD STOCK



PATTERN PAGE FOR "MAKE FIVE"  
(Dice 2 and 3)

COPY ONTO CARD STOCK



## barite



Often found in limestone or hot spring areas. Usually white or light brown. Sometimes crystallizes into rose shapes, which are popular with collectors.

## zircon



Found in nearly all igneous rocks, although in very small amounts. Because it is so hard, it is often used as a gemstone in jewelry.

## hematite



Hematite is a major ore (source) of iron. The name "hematite" comes from its blood-red color ("hema" means blood).

## cinnabar



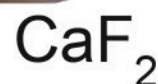
Cinnabar has a reddish color and is very dense (heavy) because of the mercury (Hg). Pure mercury is a liquid at room temperature, but it is a solid when bound to sulfur.

## cuprite



Cuprite forms cubic crystals. It is sometimes called "ruby copper" because of its color. When exposed to air it changes to  $\text{CuO}$ .

## fluorite



Fluorite is used in the production of steel. It has a glassy luster and can look similar to a quartz crystal, except for its tetragonal (4-sided) shape.

## quartz



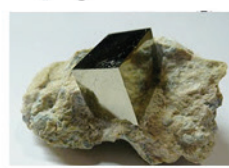
Quartz is used in electronics, as a gemstone, and in the manufacturing of glass (where it is the main component). Sand is made of very tiny pieces of quartz.

## galena



Galena is very dense (heavy) because of the lead in it. During the era of musket rifles, galena was used as the source of lead to make musket balls.

## pyrite



This mineral is often called "fool's gold" because of its golden color and shiny luster. It has no actual gold in it. It leaves a black streak, not gold.

FIRST PATTERN PAGE FOR "MAKE FIVE"

COPY ONTO CARD STOCK

## corundum



Corundum is very hard. It is so hard that it is used in industry as an abrasive (like sand paper). Blue corundum is called a sapphire and red is a ruby.

## talc



Talc is extremely soft. In fact, you can scratch it with your fingernail! Talc is the main ingredient in talcum powder (used to dry off after a shower).

## calcite



Calcite is the main ingredient in limestone. It is one of the most common minerals in the world. Caves are made of limestone.

## gypsum



Gypsum is a soft mineral. It is one of the main ingredients in plaster and plasterboard. One type of gypsum is called alabaster and was carved by ancient peoples.

## chalcopyrite



Chalcopyrite is pinkish-purple with flecks of gold. It is found wherever copper is mined. The copper can be taken out of it by using chemical processes.

## epsom salt



This mineral dissolves into water very easily. It is often used in medical treatment of wounds on hands and feet. It helps in the healing process.

## diamond/graphite



Strangely enough, both priceless diamonds and the stuff in your pencil are made of the same thing: pure carbon. The difference is how the atoms are bonded together.

SECOND PATTERN PAGE FOR "MAKE FIVE"

COPY ONTO CARD STOCK





PATTERN PAGE FOR "SYMBOL JARS"

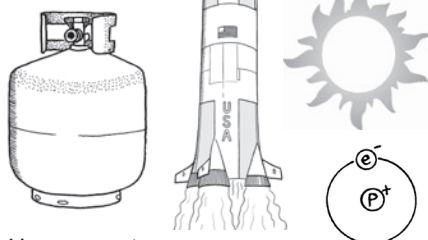
COPY ONTO CARD STOCK

# H 1

**Hydrogen**

1.0

Greek: "hydro-gen" (water-maker)



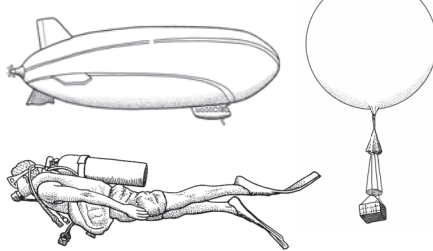
- Has no neutrons.
- Is the fuel for stars (including our sun)
- Used in rocket fuel and fuel cells
- Combines with O to make water.
- Combines with C to make natural gas.

# He 2

**Helium**

4.0

Greek: "helios" (sun)



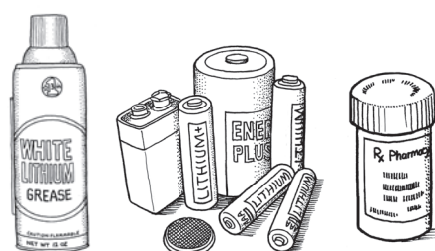
- Used in balloons, blimps and scuba diving tanks.
- Discovered in the sun in 1895 using a spectrometer.

# Li 3

**Lithium**

6.9

Greek: "lithos" (stone)



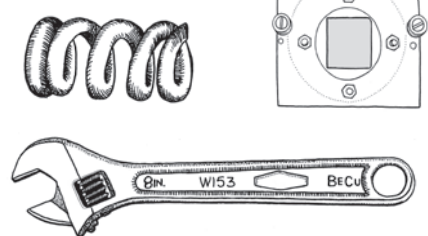
- Used in batteries, lubricants, medicines, red fireworks, and nuclear bombs.
- Is never found by itself in nature (it's always in a compound).

# Be 4

**Beryllium**

9.0

from the mineral "beryl"



- Found in emeralds.
- Is mixed with copper to make "beryllium bronze," for making tools and parts that won't create sparks.
- Used for "windows" that let x-rays pass.

# B 5

**Boron**

10.8

from the compound "borax"



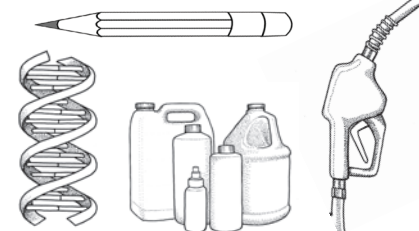
- Used to make heat-resistant glass.
- Used in boric acid antiseptic eye wash.
- Is a main ingredient in Borax® washing powder
- Used in nuclear power plants.
- Combines with white glue to make "slime"

# C 6

**Carbon**

12.0

Latin: "carbo" (charcoal)



- Diamonds, graphite (pencil "lead") and coal are all made of carbon.
- Carbon makes long chains (polymers) that are the basis of fossil fuels and plastics.
- Carbon is necessary for organic molecules found in living organisms.

# N 7

**Nitrogen**

14.0

Greek: "nitron" (the mineral saltpetre)



- Most of the air we breathe is nitrogen.
- Used in air bags in cars.
- Combines with H to make ammonia (found in some cleaning products)
- Plants need N to stay green.
- Found in gun powder.

# O 8

**Oxygen**

15.9

Greek: "oxy-gen" (acid-maker)



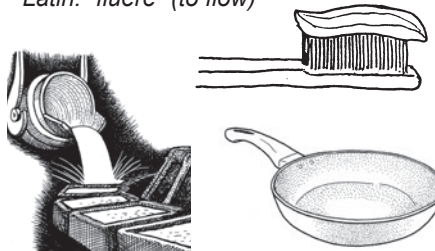
- Combines with silicon to make sand
- Necessary for respiration and combustion.
- H<sub>2</sub>O is water.
- H<sub>2</sub>O<sub>2</sub> is hydrogen peroxide.

# F 9

**Fluorine**

18.9

Latin: "fluere" (to flow)



- Combines with Ca to make fluorite.
- Is put into toothpaste to fight cavities.
- Combines with C to make Teflon.®
- Used as flux in steel making (makes hot metal flow better).

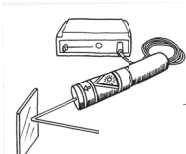
# Ne

Néon

Greek: "neo" (new)

# 10

20.1



- Used in neon lights and lasers.
- Neon doesn't bond to other elements.

# Na

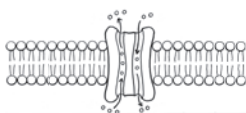
Sodium

from soda ash



# 11

22.9



- Combines with Cl to make table salt.
- Used in street lights.
- Found in bleach,  $\text{NaClO}$ .
- Found in sodium/potassium pumps in cell membranes.

# Mg

Magnesium

from Magnesia, in Greece



# 12

24.3



- Used in white sparklers.
- Found in Epsom salt,  $\text{MgSO}_4$ .
- Is the central atom in chlorophyll (the green molecule in leaves).
- Used to make fire starters.

# Al

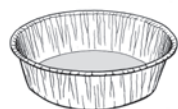
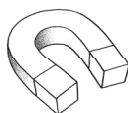
Aluminum

from the compound "alumina"



# 13

26.9



- Used to build bicycles and airplanes.
- $\text{Al}_2\text{O}_3$  is the mineral (gem) corundum.
- Used for foil, and for beverage cans.
- Used in AlNiCo magnets.

# Si

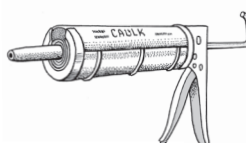
Silicon

Latin: "silex" (hard stone, boulder)



# 14

28.0

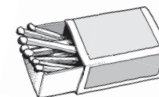


- $\text{SiO}_2$  is quartz (which can form sand).
- Used to make microchips.
- Used to make silicon rubber products like caulk and flexible baking dishes.

# P

Phosphorus

Greek: "phosphoros" (bringer of light)



# 15

30.9

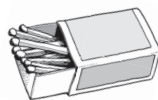


- Found in matches and some cleaners.
- Necessary for strong bones and teeth.
- Phosphoric acid is often in fizzy drinks.
- Is an ingredient in many plant fertilizers.

# S

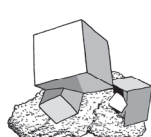
Sulfur

Latin: "sulfur" (stone that burns)



# 16

32.0

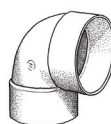


- Used to make matches.
- Used to vulcanize rubber.
- Volcanoes produce gases containing sulfur.
- Makes skunks and garlic stinky.
- Combines with Fe to make the mineral pyrite.

# Cl

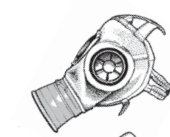
Chlorine

Greek: "kloros" (light green)



# 17

35.4



- Bonds with Na to make table salt.
- Used to disinfect swimming pools.
- Is an ingredient in PVC plastics.
- Found in bleach,  $\text{NaClO}$ .
- In pure form is a poisonous green gas.

# Ar

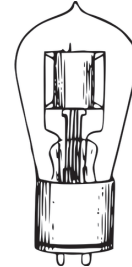
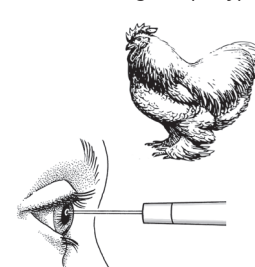
Argon

Greek: "argos" (lazy)



# 18

39.9



- Replaces air in lightbulbs.
- Used to make lasers for eye surgery.
- Used to butcher chickens.
- Doesn't form any molecules.

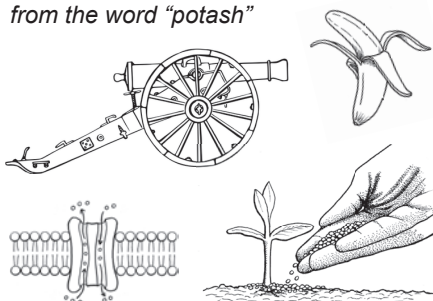


# K 19

**Potassium**

**39.0**

from the word "potash"



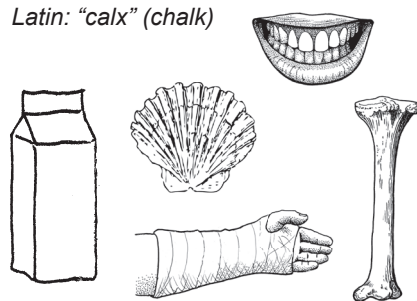
- Used in fertilizers.
- Is an ingredient in gun powder.
- Bananas contain a lot of potassium.
- Is found in mineral orthoclase feldspar.

# Ca 20

**Calcium**

**40.0**

Latin: "calx" (chalk)



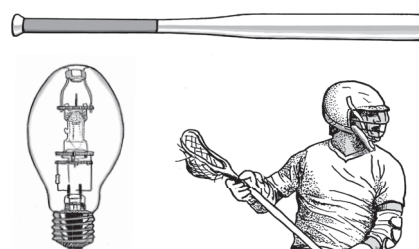
- Found in chalk, limestone and plaster.
- Needed for strong bones and teeth.
- Milk and broccoli have lots of calcium.
- Seashells are made with  $\text{CaCO}_3$ .

# Sc 21

**Scandium**

**44.9**

named after Scandinavia



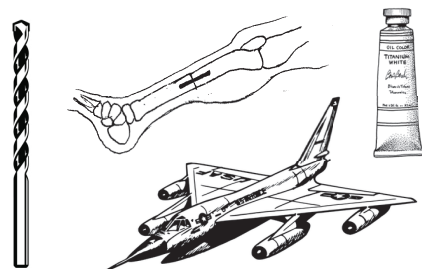
- Used in high intensity light bulbs.
- Used in large television screens.
- Alloys containing Sc are used to make sports equipment.

# Ti 22

**Titanium**

**47.9**

named after the Greek Titan gods



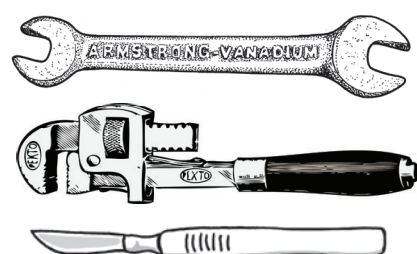
- Used to repair bones.
- Used to make white pigment for paint.
- Alloys are used to make airplane engines, drill bits and tools.

# V 23

**Vanadium**

**50.9**

after the Scandinavian goddess Vanadis



- Added to steel to make very durable and corrosion-resistant tools, springs and engine parts

# Cr 24

**Chromium**

**51.9**

Greek: "chroma" (color)



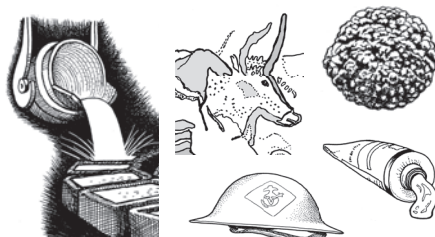
- Gives rubies their red color.
- Used to make red, green and yellow paint.
- Used as a shiny coating for metals.
- Added to steel to make it "stainless."

# Mn 25

**Manganese**

**54.9**

Latin: "magnes" (magnetic)



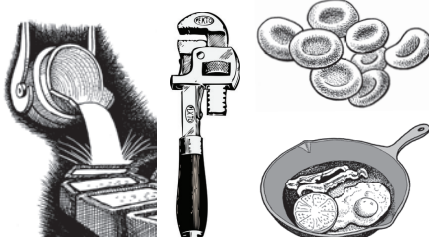
- Used to remove impurities from steel.
- Manganese nodules are on ocean floor.
- Found in vitamin B1.
- Is the "Mn" in  $\text{YInMn}$  blue pigment.
- Found in pigments in cave paintings.

# Fe 26

**Iron**

**55.8**

from Old English "iren"



- Discovered in ancient times.
- Used in steel and in magnets.
- Found in red blood cells and in rust.
- Used to make cast iron cooking pans.
- Red blood cells are red due to iron.

# Co 27

**Cobalt**

**58.9**

German "kobald" (evil gnomes)



- Used in  $\text{AlNiCo}$  magnets.
- Used in making drill bits and razors.
- Can be used to color glass deep blue. or to make blue glaze/paint for pottery.

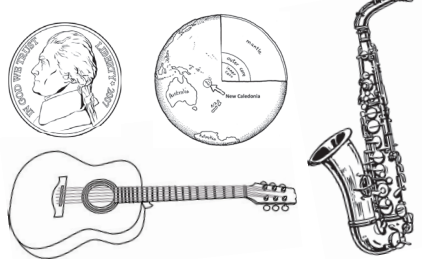


# Ni 28

**Nickel**

**58.7**

German: "Nickel" (Satan)



- Core of earth likely made of Ni and Fe.
- Used to make coins and utensils.
- Used as coating for keys of musical
- Used to make AlNiCo magnets.
- Guitar strings made of nickel and steel.

# Cu 29

**Copper**

**63.5**

Latin: "Cuprum" (from Cyprus)



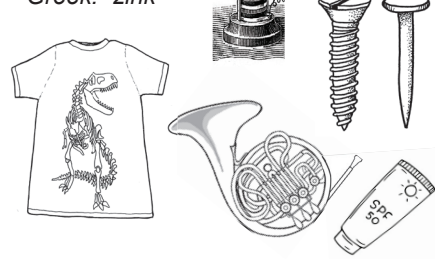
- Used for coins, wires and pipes.
- The Statue of Liberty is made of copper.
- Copper mixed with zinc makes brass.
- Copper mixed with tin makes bronze.

# Zn 30

**Zinc**

**65.4**

Greek: "zink"



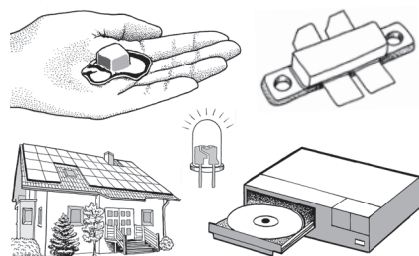
- Used for galvanizing (protecting)
- Was used to make voltaic pile battery.
- Zinc sulfide glows in the dark.
- Zinc and copper make brass.
- Sunscreens can contain Zn compounds

# Ga 31

**Gallium**

**69.7**

Latin: "Gallia" (France)



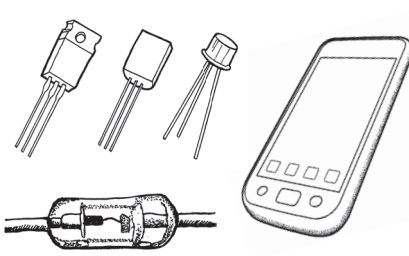
- Gallium arsenide is used in LEDs, lasers, and in Blu-ray players.
- Used in electronic devices.
- Gallium arsenide is in solar panels.

# Ge 32

**Germanium**

**72.6**

Latin: "Germania" (Germany)



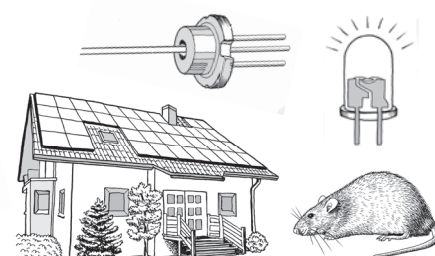
- Is a semi-conductor and therefore is used in transistors and diodes.
- Used in photographic lenses.
- Used in infrared-sensing devices.

# As 33

**Arsenic**

**74.9**

Latin: "arsenicum" (a pigment)



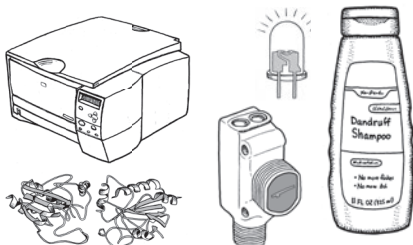
- Famous for its use as a poison.
- Gallium arsenide is in solar panels.
- Used in lasers and LED's.
- In the past, was used for green pigment.

# Se 34

**Selenium**

**78.9**

Greek: "selene" (moon)



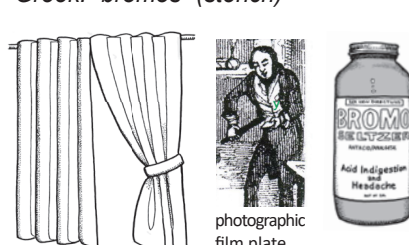
- Used in photocopiers because it conducts electricity in the presence of light.
- Used in solar panels and in light meters.
- Selenium used as an anti-oxidant in our bodies, protecting us from cellular damage.
- A key ingredient in anti-dandruff shampoo.

# Br 35

**Bromine**

**79.9**

Greek: "bromos" (stench)



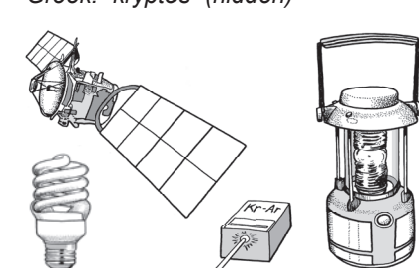
- Bromine is in the purple ink taken out of murex mollusks (royal purple)
- Used to make fire-resistant fabrics.
- Was used in photographic film.
- Was key ingredient in Bromo-Seltzer.®

# Kr 36

**Krypton**

**83.8**

Greek: "kryptos" (hidden)



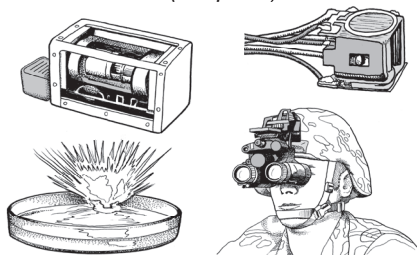
- Used in fluorescent bulbs, and for bulbs that need to be extremely bright.
- Used in UV lasers and in atomic clocks.
- Was used as propellant in satellites.

# Rb 37

**Rubidium**

**85.5**

Latin: "rubidus" (deep red)



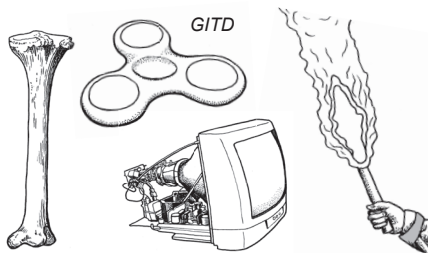
- Will burn in water (red flames).
- Used as a "scavenger" in vacuum tubes.
- Used in magnetometers and night vision goggles.
- Used in small atomic clocks.

# Sr 38

**Strontium**

**87.6**

after the Scottish village of Strontia



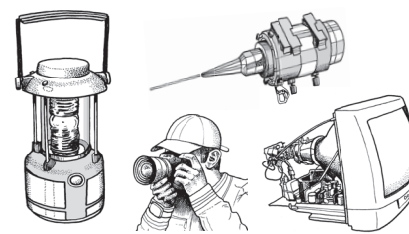
- Used in fireworks and flares.
- Sr in old bones is used by archaeologists.
- Was used in CRT television screens.
- Strontium aluminate glows in the dark.

# Y 39

**Yttrium**

**88.9**

after the Swedish town of Ytterby



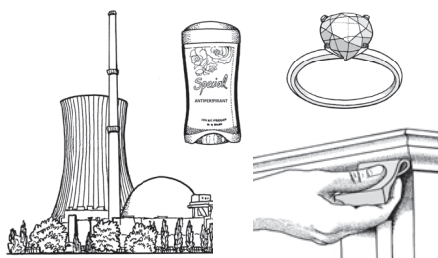
- Used in mantles for gas lanterns.
- Used in YAG (yttrium garnet) lasers.
- Made red color in CRT televisions.
- Is the "Y" in YInMn blue p
- Used in glass for specialty lenses.

# Zr 40

**Zirconium**

**91.2**

Arabic: "zargun" (gold color)



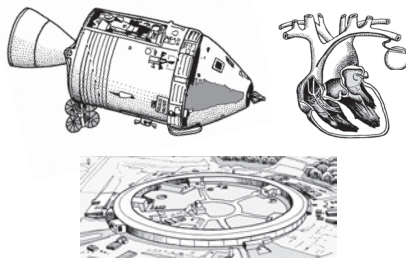
- Made into gemstones.
- Found in antiperspirants (along with Al)
- Used for tubes in nuclear power plants •
- Used to make abrasives (sand paper)

# Nb 41

**Niobium**

**92.9**

named after the Greek goddess Niobe



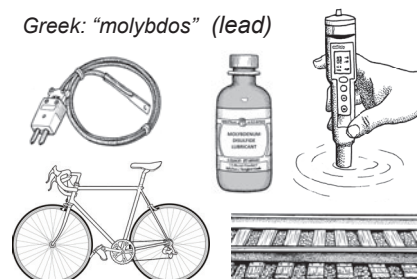
- Used in welding rods and cutting tools.
- Used in alloys for rocket engines.
- Niobium jewelry is iridescent.
- Nb wired used in particle accelerators.

# Mo 42

**Molybdenum**

**95.9**

Greek: "molybdos" (lead)



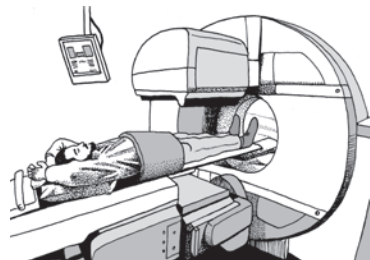
- Used for filaments in heaters.
- Added to steel ("Chromoly" steel) used for things that must be durable.
- Used in water quality gauges.
- MoS<sub>2</sub> is used as a dry lubricant.

# Tc 43

**Technetium**

**99.0**

Greek: "teknetos" (artificial)



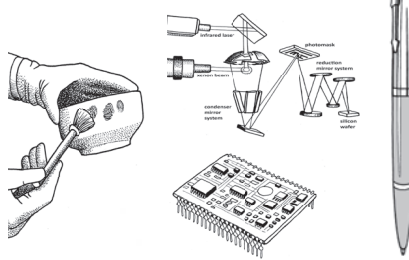
- Is radioactive.
- Not found in nature. Must be made in a nuclear laboratory.
- Used in medical procedures.

# Ru 44

**Ruthenium**

**101.1**

Latin: "Ruthenia" (Russia)



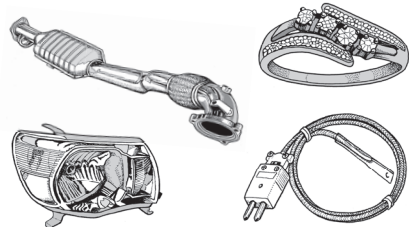
- Used to make resistor chips for electronic devices
- Was used to make tips for fountain pens.
- Used for lifting fingerprints.

# Rh 45

**Rhodium**

**102.9**

Greek: "rhodon" (rose)



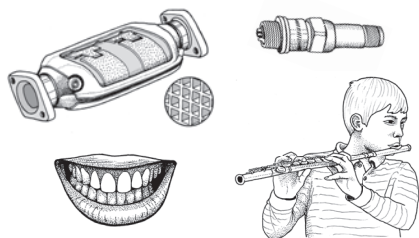
- Used in catalytic converters in cars.
- Used in headlight reflectors.
- Makes jewelry look extra shiny.
- Combined with **Pt** and **Pd** to make spark plugs and electrodes.



# Pd 46

**Palladium** 106.4

*named after the asteroid Pallas*



- Used in catalytic converters in cars.
- Found in airplane spark plugs.
- Used to make high quality flutes.
- Used in jewelry and dentistry.
- Used with silver for capacitors.

# Ag 47

**Silver** 107.8

*Anglo-Saxon: "soilful" (silver)  
Symbol from Latin "argentum"*

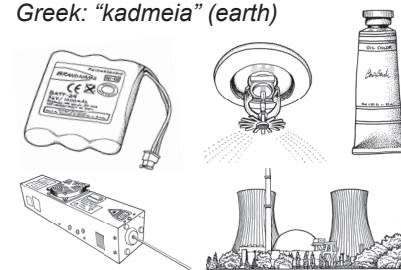


- Used to make coins, jewelry, mirrors, silverware, photographic film.
- Used to make anti-bacterial bandages.
- Bang snaps use silver fulminate.
- Found, with silver, in capacitors.

# Cd 48

**Cadmium** 112.4

*Greek: "kadmeia" (earth)*

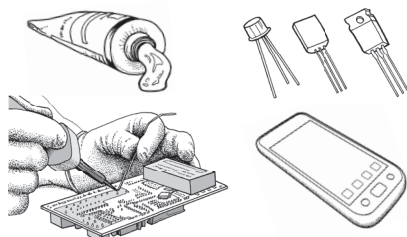


- Used in rechargeable batteries.
- Used in some lasers.
- Makes yellow and red pigments.
- Used (with Bi) for sprinkler fuses.
- Absorbs neutrons in nuclear reactors.

# In 49

**Indium** 114.8

*Latin: "indicum" (indigo blue)*



- Used in transistors and solar cells.
- Used in soldering.
- Indium-tin-oxide, ITO, is used as a conductive surface for touch screens.
- Is the "In" in YInMn blue pigment.

# Sn 50

**Tin** 118.7

*Latin: "stannum" (tin)*

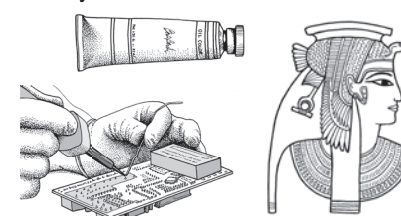


- Is an ingredient of pewter.
- Is mixed with copper to make bronze.
- Indium-tin-oxide, ITO, is used as a conductive surface for touch screens.
- Metal toys used to be made of tin.

# Sb 51

**Antimony** 121.7

*Greek: "anti-monos" (not alone)  
Symbol comes from "stibnium"*

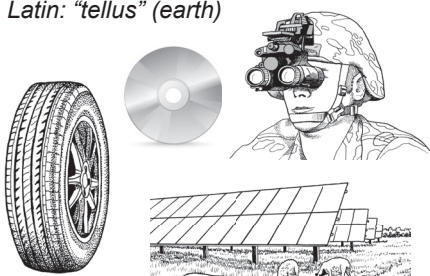


- Used in solder.
- Key ingredient in Naples yellow paint.
- Found in safety matches, and is added to fabrics to make the fire-resistant.
- Ancients used it in eye-liner cosmetics.

# Te 52

**Tellurium** 127.6

*Latin: "tellus" (earth)*

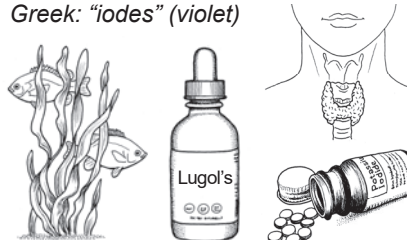


- Can replace sulfur in vulcanization process.
- HgCdTe is used to sense infrared, used in military night-vision equipment
- CdTe is used in solar panels.
- Used to make Blu-ray discs.

# I 53

**Iodine** 126.9

*Greek: "iodes" (violet)*

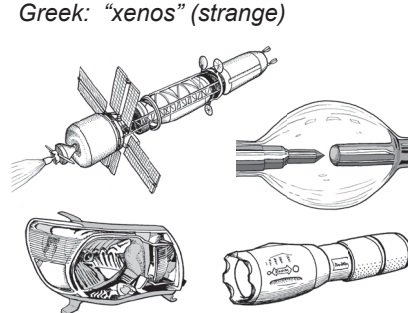


- Used as a disinfectant.
- Used in halogen lamps, ink pigments and photographic film.
- Our thyroid glands need iodine.
- Silver iodide is used to make clouds rain.
- Found naturally in seaweed.

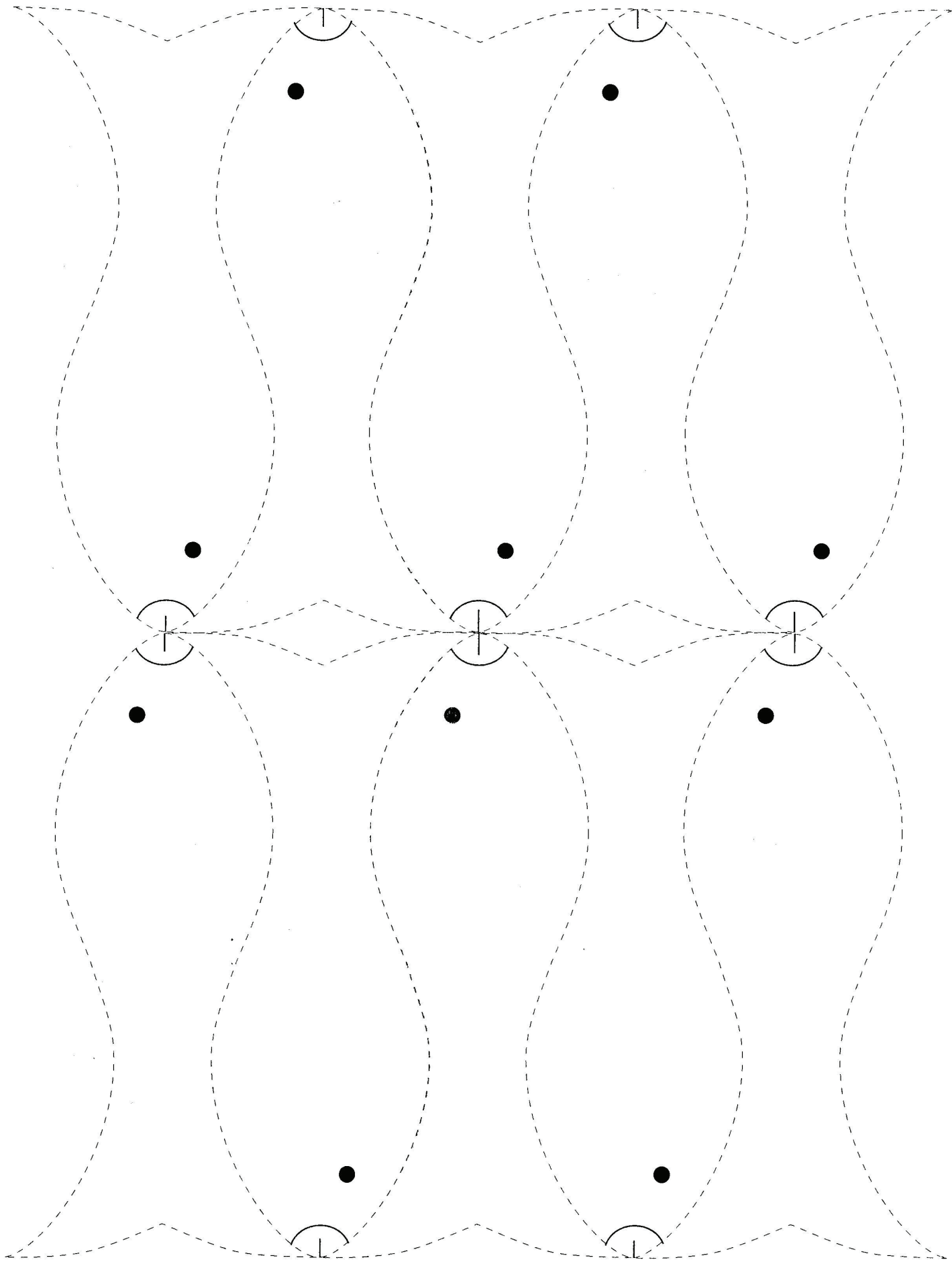
# Xe 54

**Xenon** 131.3

*Greek: "xenos" (strange)*

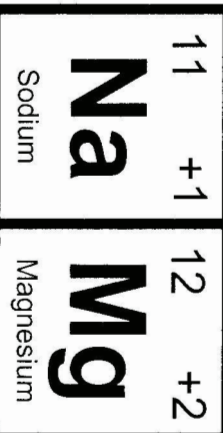
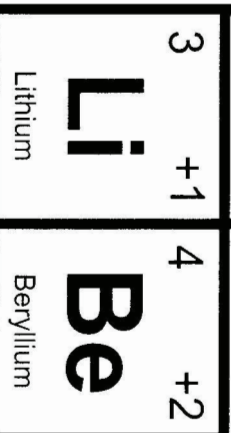
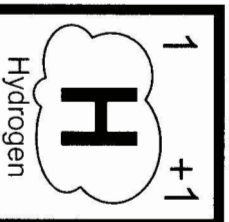


- Used in camera flash bulbs, strobe lights, and other high intensity bulbs.
- Was used as propellant for satellites.





# Start



19	+1	20	+2	21	+3	22	+4	23	+5	24	+6	25	+7	26	+3	27	+3
<b>K</b>		<b>Ca</b>		<b>Sc</b>		<b>Ti</b>		<b>V</b>		<b>Cr</b>		<b>Mn</b>		<b>Fe</b>		<b>Co</b>	
Potassium		Calcium		Scandium		Titanium		Vanadium		Chromium		Manganese		Iron		Cobalt	
37	+1	38	+2	39	+3	40	+4	41	+5	42	+6	43	+7	44	+3	45	+3
<b>Rb</b>		<b>Sr</b>		<b>Y</b>		<b>Zr</b>		<b>Nb</b>		<b>Mo</b>		<b>Tc</b>		<b>Ru</b>		<b>Rh</b>	
Rubidium		Strontium		Yttrium		Zirconium		Niobium		Molybdenum		Technetium		Ruthenium		Rhodium	

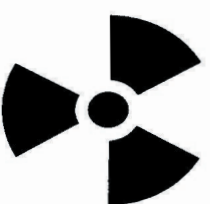
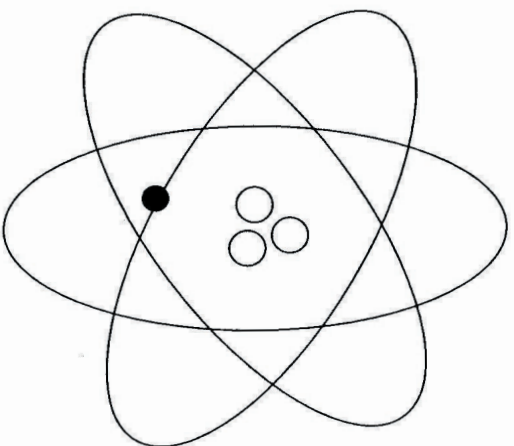
# THE Periodic Table Game

Liquid or Gas at Room Temperature > Roll Again

Radioactive > Put up Shields on Either Side



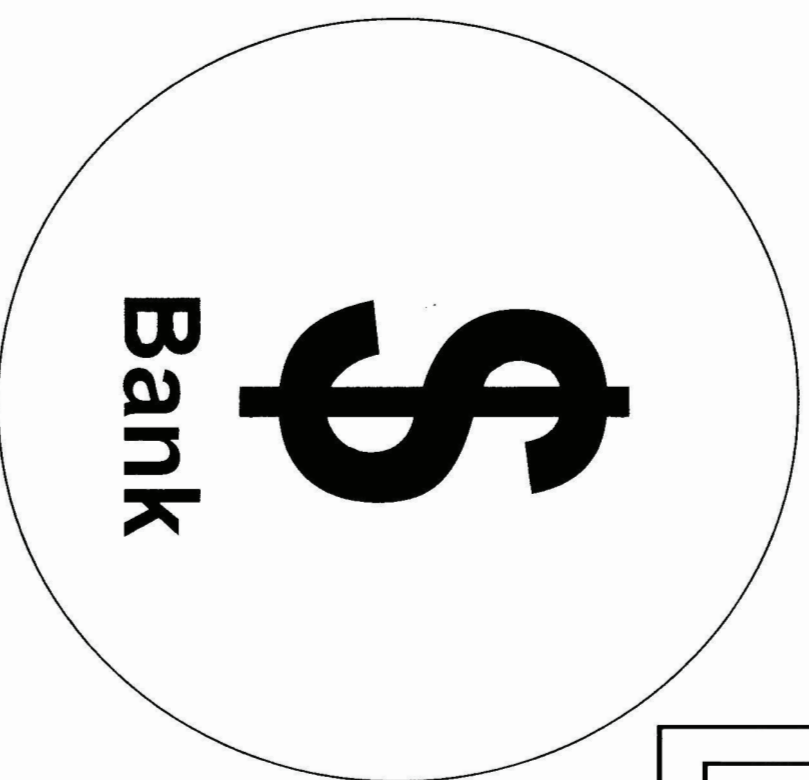
Named After Person or Place > 3 Extra Pennies if You Name It!



# Radioactive Shields

28	+3	<b>Ni</b>	Nickel	29	+2	<b>Cu</b>	Copper	30	+2	<b>Zn</b>	Zinc
46	+4	<b>Pd</b>	Palladium	47	+1	<b>Ag</b>	Silver	48	+2	<b>Cd</b>	Cadmium
5	+3	<b>B</b>	Boron	6	+4	<b>C</b>	Carbon	7	-3	<b>N</b>	Nitrogen
13	+3	<b>Al</b>	Aluminum	14	+4	<b>Si</b>	Silicon	15	-3	<b>P</b>	Phosphorus
31	+3	<b>Ga</b>	Gallium	32	+4	<b>Ge</b>	Germanium	33	-3	<b>As</b>	Arsenic
49	+3	<b>In</b>	Indium	50	+4	<b>Sn</b>	Tin	51	-3	<b>Sb</b>	Antimony
52	-2	<b>Te</b>	Tellurium	53	-1	<b>I</b>	Iodine	34	-2	<b>Se</b>	Selenium
35	-1	<b>Br</b>	Bromine	36	0	<b>Kr</b>	Krypton	16	-2	<b>S</b>	Sulfur
17	-1	<b>Cl</b>	Chlorine	18	0	<b>Ar</b>	Argon	9	-1	<b>F</b>	Fluorine
10	0	<b>Ne</b>	Neon	54	0	<b>Xe</b>	Xenon	2	0	<b>He</b>	Helium

55	+1	<b>Cs</b> Cesium	56	+2	<b>Ba</b> Barium	57	+3	<b>La</b> Lanthanum	72	+4	<b>Hf</b> Hafnium	73	+5	<b>Ta</b> Tantalum	74	+6	<b>W</b> Tungsten	75	+7	<b>Re</b> Rhenium	76	+3	<b>Os</b> Osmium	77	+4	<b>Ir</b> Iridium
87	+1	<b>Fr</b> Francium	88	+2	<b>Ra</b> Radium	89	+3	<b>Ac</b> Actinium	104	+4	<b>Rf</b> Rutherfordium	105		<b>Db</b>	106		<b>Sg</b>	107		<b>Bh</b>	108		<b>Hs</b>	109		<b>Mt</b>



58	+3	<b>Ce</b> Cerium	59	+3	<b>Pr</b> Praseodymium	60	+3	<b>Nd</b> Neodymium	61	+3	<b>Pm</b> Promethium	62	+3	<b>Sm</b> Samarium
90	+4	<b>Th</b> Thorium	91	+5	<b>Pa</b> Protactinium	92	+6	<b>U</b> Uranium	93	+6	<b>Np</b> Neptunium	94	+6	<b>Pu</b> Plutonium

78	+4	<b>Pt</b>	Platinum
79	+3	<b>Au</b>	Gold
80	+2	<b>Hg</b>	Mercury
81	+3	<b>Tl</b>	Thallium
82	+4	<b>Pb</b>	Lead
83	-3	<b>Bi</b>	Bismuth
84	-2	<b>Po</b>	Polonium
85	-1	<b>At</b>	Astatine
86	0	<b>Rn</b>	Radon

110	<b>Ds</b>
111	<b>Rg</b>
112	<b>Cn</b>
113	<b>Nh</b>
114	<b>Fl</b>
115	<b>Mc</b>
116	<b>Lv</b>
117	<b>Ts</b>
118	<b>Og</b>

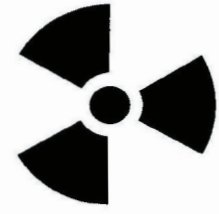
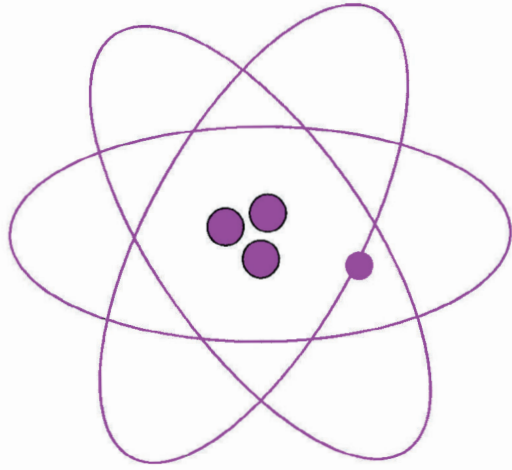
Go to 72 Hf

63	+3	<b>Eu</b>	Europium
64	+3	<b>Gd</b>	Gadolinium
65	+3	<b>Tb</b>	Terbium
66	+3	<b>Dy</b>	Dysprosium
67	+3	<b>Ho</b>	Holmium
68	+3	<b>Er</b>	Erbium
69	+3	<b>Tm</b>	Thulium
70	+3	<b>Yb</b>	Ytterbium
71	+3	<b>Lu</b>	Lutetium
95	+6	<b>Am</b>	Americium
96	+4	<b>Cm</b>	Curium
97	+4	<b>Bk</b>	Berkelium
98	+4	<b>Cf</b>	Californium
99	+3	<b>Es</b>	Einsteinium
100	+3	<b>Fm</b>	Fermium
101	+3	<b>Md</b>	Mendelevium
102	+3	<b>No</b>	Nobelium
103	+3	<b>Lr</b>	Lawrencium

Go to 104 Rf



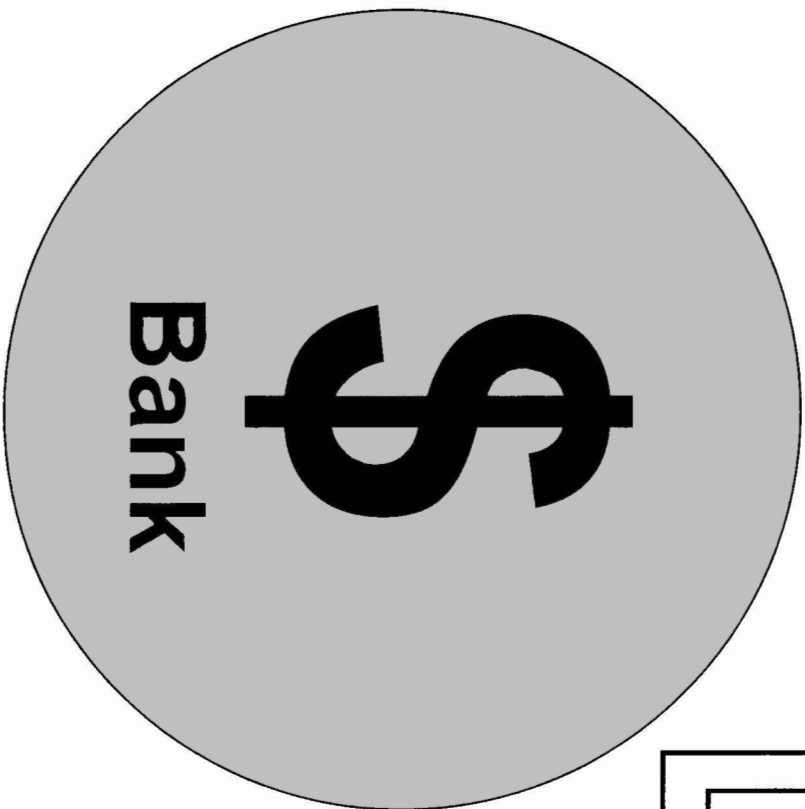




# Radioactive Shields

28	+3	Ni	Nickel	29	+2	Cu	Copper	30	+2	Zn	Zinc	36	0	Kr	Krypton
46	+4	Pd	Palladium	47	+1	Ag	Silver	48	+2	Cd	Cadmium	54	0	Xe	Xenon
51	-3	Sb	Antimony	50	+4	Sn	Tin	52	-2	Te	Tellurium	53	-1	I	Iodine
33	-3	As	Arsenic	32	+4	Ge	Germanium	34	-2	Se	Selenium	35	-1	Br	Bromine
15	-3	P	Phosphorus	14	+4	Si	Silicon	16	-2	S	Sulfur	17	-1	Cl	Chlorine
7	-3	N	Nitrogen	6	+4	C	Carbon	8	-2	O	Oxygen	9	-1	F	Fluorine
13	+3	Al	Aluminum	13	+3	Al	Aluminum	14	+4	Si	Silicon	10	0	Ne	Neon
5	+3	B	Boron	6	+4	C	Carbon	7	-3	N	Nitrogen	2	0	He	Helium

55 <b>Cs</b> Cesium	56 <b>Ba</b> Barium	57 <b>La</b> Lanthanum	72 <b>Hf</b> Hafnium	73 <b>Ta</b> Tantalum	74 <b>W</b> Tungsten	75 <b>Re</b> Rhenium	76 <b>Os</b> Osmium	77 <b>Ir</b> Iridium
87 <b>Fr</b> Francium	88 <b>Ra</b> Radium	89 <b>Ac</b> Actinium	104 <b>Rf</b> Rutherfordium	105 <b>Db</b>	106 <b>Sg</b>	107 <b>Bh</b>	108 <b>Hs</b>	109 <b>Mt</b>



58 <b>Ce</b> Cerium	59 <b>Pr</b> Praseodymium	60 <b>Nd</b> Neodymium	61 <b>Pm</b> Promethium	62 <b>Sm</b> Samarium
90 <b>Th</b> Thorium	91 <b>Pa</b> Protactinium	92 <b>U</b> Uranium	93 <b>Np</b> Neptunium	94 <b>Pu</b> Plutonium

78 +4 <b>Pt</b> Platinum	79 +3 <b>Au</b> Gold	80 +2 <b>Hg</b> Mercury	81 +3 <b>Tl</b> Thallium	82 +4 <b>Pb</b> Lead	83 -3 <b>Bi</b> Bismuth	84 -2 <b>Po</b> Polonium	85 -1 <b>At</b> Astatine	86 0 <b>Rn</b> Radon
110 <b>Ds</b>	111 <b>Rg</b>	112 <b>Cn</b>	113 <b>Nh</b>	114 <b>Fl</b>	115 <b>Mc</b>	116 <b>Lv</b>	117 <b>Ts</b>	118 <b>Og</b>

Go to 72 Hf →

63 +3 <b>Eu</b> Europium	64 +3 <b>Gd</b> Gadolinium	65 +3 <b>Tb</b> Terbium	66 +3 <b>Dy</b> Dysprosium	67 +3 <b>Ho</b> Holmium	68 +3 <b>Er</b> Erbium	69 +3 <b>Tm</b> Thulium	70 +3 <b>Yb</b> Ytterbium	71 +3 <b>Lu</b> Lutetium
95 +6 <b>Am</b> Americium	96 +4 <b>Cm</b> Curium	97 +4 <b>Bk</b> Berkelium	98 +4 <b>Cf</b> Californium	99 +3 <b>Es</b> Einsteinium	100 +3 <b>Fm</b> Fermium	101 +3 <b>Md</b> Mendelevium	102 +3 <b>No</b> Nobelium	103 +3 <b>Lr</b> Lawrencium

Go to 104 Rf →



## 2) CRAFT: Make a Periodic Table pillowcase

You will need: copies of the following pattern pages, clear tape, a blank pillowcase (white or a very light pastel color is best), fabric markers, glow-in-the-dark paint (if possible), and some pins to hold the pattern in place (and an iron if the instructions on your fabric markers say to use one)

### What to do:

1) Copy the pattern pages onto regular paper (no need for card stock). Tape the four pages together so that they form a blank Periodic Table. Put this inside the pillowcase. You should be able to see the black lines right through the fabric. Adjust the pattern so that it is placed in the middle of the pillowcase, and pin it in place.

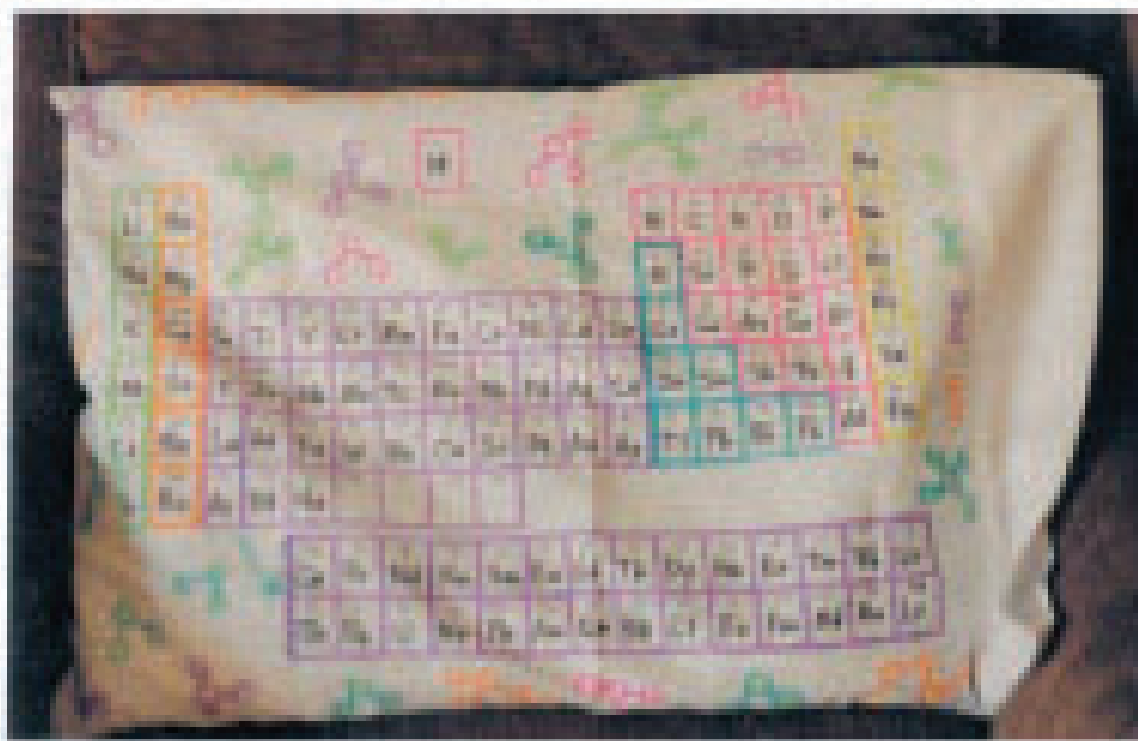
NOTE: A few key numbers are given as a guide, to prevent major mistakes such as forgetting to jump down to the lanthanide and actinide series down below, or going top to bottom instead of left to right.

2) Use the fabric markers to trace over the squares. Color code the families. You don't have to use the color code shown here. You can decide what color to make each family. (If you want to add more elements, after 109, you are welcome to add them. These are what I call the "extremely silly elements" because they really don't exist. A few atoms blink in and out of existence for a millionth of a second. But you are welcome to add them to that bottom row if you want to.)

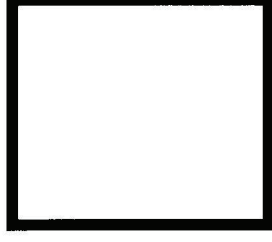
3) Write in the symbol for each element and its atomic number.

4) FUN EXTRA FEATURE: You could put glow-in-the-dark paint on the radioactive elements. GITD paint is easily obtained from any craft store and is not expensive. Look at your Periodic Table game (or find a Periodic Table on the Internet) to see which elements are radioactive. (Don't forget Technetium!)

5) Follow any ironing or washing instructions that come with your fabric markers.



Upper Left



# Hydrogen

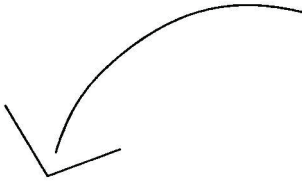
[illegible]

Upper Right

5	13			




55							
87		89	104				

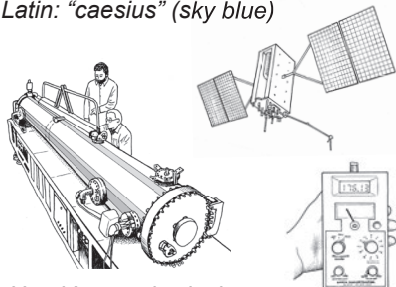


58					
90					

Lower Left

**Cs**  **55**  
Cesium 132.9

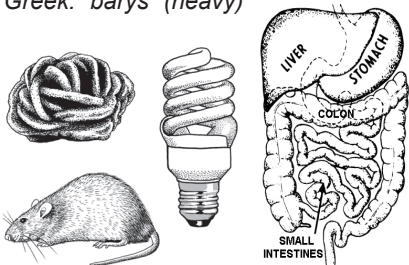
Latin: "caesius" (sky blue)



- Used in atomic clocks.
- GPS satellite use these atomic clocks.
- Used as a "scavenger" (collector) inside vacuum tubes.
- Found in some magnetometers.

**Ba**  **56**  
Barium 137.3

Greek: "barys" (heavy)



- Used for X-rays of digestive systems.
- Used in fireworks (green color).
- Found in mineral barite (desert rose).
- BaO is in electrodes in fluorescent lights.
- Barium carbonate was used to kill rats.

**La** **57**  
Lanthanum 138.9

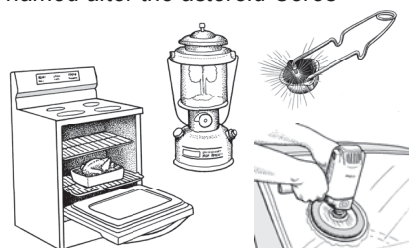
Greek: "lanthanein" (to lie hidden)



- Used in telescope and camera lenses.
- Used for electrodes in high intensity lights and in mantles for gas lanterns.
- Used in spark-making devices
- Found in some algae-killing solutions.

**Ce** **58**  
Cerium 140.1

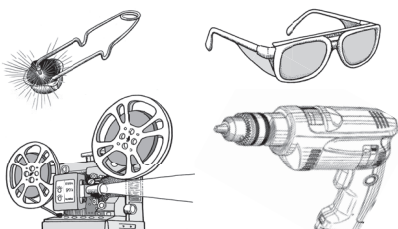
named after the asteroid Ceres



- Used in mantles for gas lanterns.
- Used in self-cleaning ovens.
- Used in sparking devices.
- Cerium oxide is used to polish glass.
- Extracted from monazite sand.

**Pr** **59**  
Praseodymium 140.9

Greek: "prasios-didymos" (green twin)



- Used in dymium glasses for welders.
- Used in bulbs for movie projectors.
- Used in magnets in electric tools.
- Sometimes found in sparking devices.
- Extracted from monazite sand.

**Nd**  **60**  
Neodymium 144.2

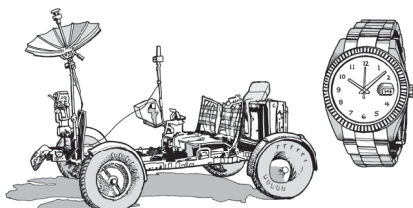
Greek: "neos-didymos" (new twin)



- Used in dymium glasses for welders.
- Used to make strong magnets found in headphones and other electronic devices.
- Used to color glass (blue, green, purple).
- Extracted from monazite sand.

**Pm**  **61**  
Promethium 147.0

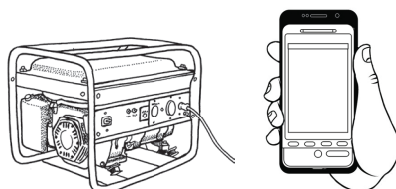
named after Greek god Prometheus



- Is a synthetic element made in nuclear reactors or cyclotrons.
- Was used to make glow-in-the-dark paint for the Apollo lunar rover.
- Replaced radium to make glowing paint for watches and clocks.

**Sm** **62**  
Samarium 150.3

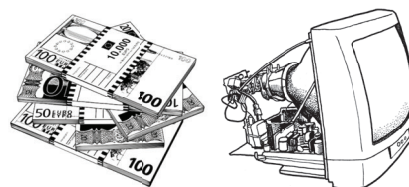
named after the mineral "samarskite" which was named for Col. Samarski, a Russian army engineer



- Used in magnets for MRI machines, computers and cell phones.
- Samarium-cobalt magnets found in heavy-duty generators.
- Extracted from monazite sand.

**Eu**  **63**  
Europium 151.9

named after Europe

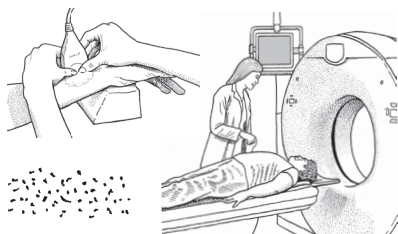


- Used to make red color in old CRT televisions.
- Used in fluorescent bulbs.
- Used to make fluorescent marks on Euros to prevent counterfeiting.
- Extracted from monazite sand.

# Gd 64

**Gadolinium** 157.2

*named for chemist Johann Gadolin*

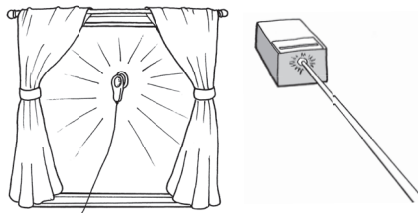


- Used in fluorescent bulbs (green light).
- Radioactive isotopes used in bone scans.
- Used as tracer in MRI scans.
- Extracted from monazite sand.

# Tb 65

**Terbium** 158.9

*named after Swedish village of Ytterby*

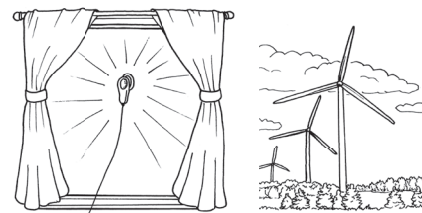


- Used to make Terfenol-D, which can turn any surface into a speaker.
- Used in fluorescent bulbs (glows green).
- Used in some green lasers.
- Extracted from monazite sand.

# Dy 66

**Dysprosium** 162.5

*Greek: "dysprositos" (difficult to obtain)*

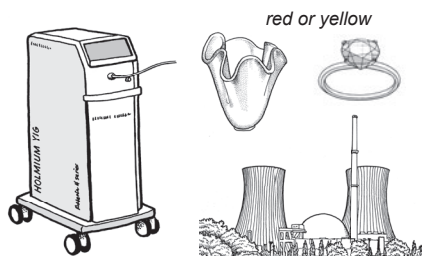


- Used to make Terfenol-D, which can turn any surface into a speaker.
- Used in wind generator magnets.
- Found in high intensity light bulbs.
- Extracted from monazite sand.

# Ho 67

**Holmium** 164.9

*named for Stockholm, Sweden*

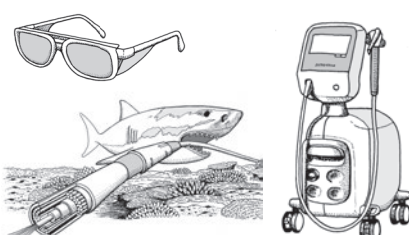


- Used in eye-safe medical lasers.
- Used to color glass red or yellow.
- Absorbs neutrons in nuclear reactors.
- Extracted from monazite sand.

# Er 68

**Erbium** 167.3

*named after Swedish village of Ytterby*

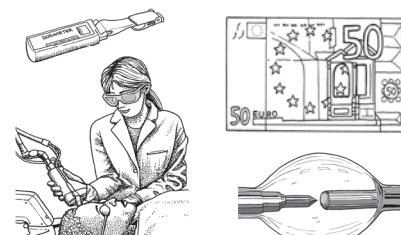


- Erbium glasses protect a patient's eyes during laser surgery.
- Used for coloring glass pink.
- Used to make medical lasers.
- Used to make relays in fiberoptic cables.

# Tm 69

**Thulium** 168.9

*Thule is the ancient name for Scandinavia*

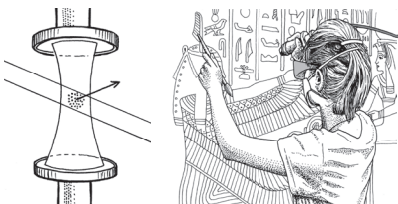


- Used in medical lasers.
- Makes fluorescent blue strip in Euro notes.
- Used in some radiation dosimeters.
- Used in some arc light bulbs.
- Extracted from monazite sand.

# Yb 70

**Ytterbium** 173.0

*named after Swedish village of Ytterby*

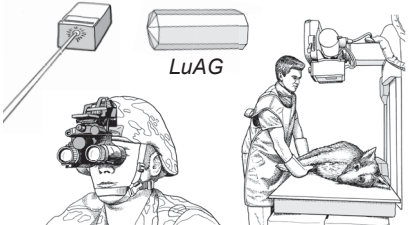


- Used in precision lasers that can clean ancient artifacts and famous paintings.
- Is added to steel to improve strength.
- Yb fluoride can be used in dental fillings.
- Yb atomic clocks use suspended atoms.
- Extracted from monazite sand.

# Lu 71

**Lutetium** 174.9

*Lutetia is the ancient name for Paris*

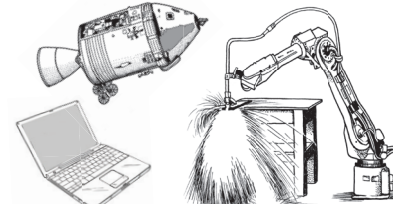


- $\text{LaTaO}_4$  is used in x-ray machines.
- Alloys of Lu are used to refine petroleum.
- Is the "Lu" in LuAG crystal lasers.
- Used in infrared sensing night-vision.
- Extracted from monazite sand.

# Hf 72

**Hafnium** 178.5

*Hafnia is the ancient name for Copenhagen*



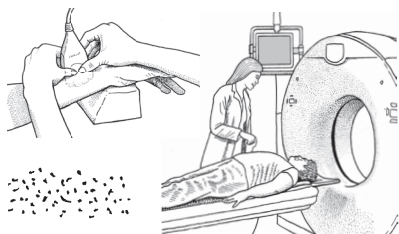
- Usually found with zirconium.
- Hf alloys were used to make nozzles of Apollo lunar module.
- Used for metal tips on plasma cutters.
- Enabled a critical step in learning to make smaller computer chips.



# Gd 64

**Gadolinium** 157.2

*named for chemist Johann Gadolin*

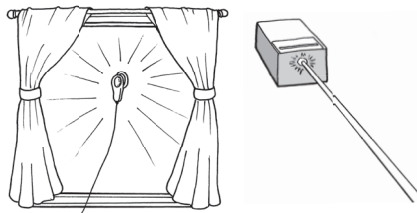


- Used in fluorescent bulbs (green light).
- Radioactive isotopes used in bone scans.
- Used as tracer in MRI scans.
- Extracted from monazite sand.

# Tb 65

**Terbium** 158.9

*named after Swedish village of Ytterby*

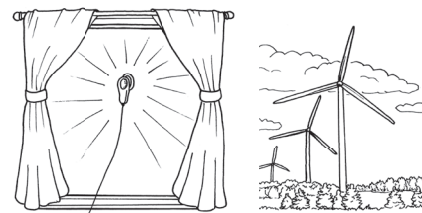


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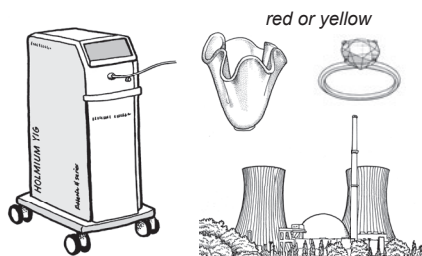


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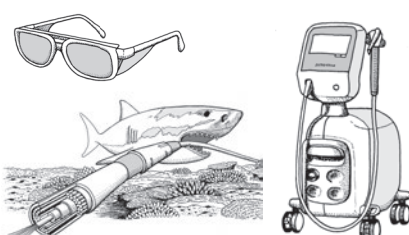


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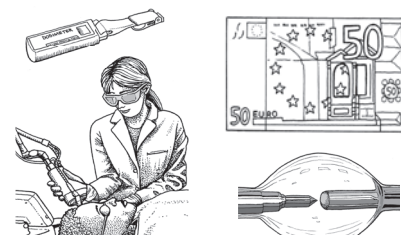


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- Used to make relays in fiberoptic cables.

# Tm 69

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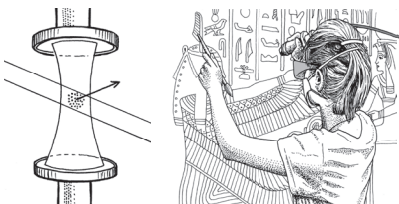


- Used in medical lasers.
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- Used in some arc light bulbs.
- Extracted from monazite sand.

# Yb 70

**Ytterbium** 173.0

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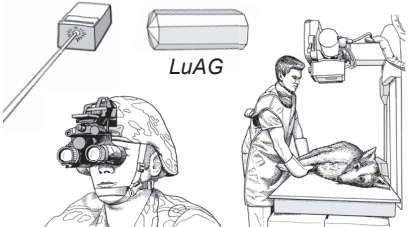


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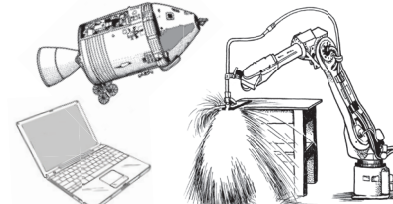


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# Hf 72

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*Hafnia is the ancient name for Copenhagen*

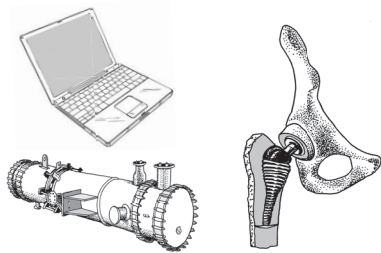


- Usually found with zirconium.
- Hf alloys were used to make nozzles of Apollo lunar module.
- Used for metal tips on plasma cutters.
- Enabled a critical step in learning to make smaller computer chips.

# Ta 73

**Tantalum** 180.9

*named after the Greek god Tantalus*

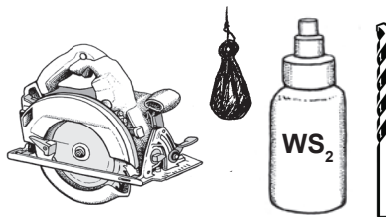


- Used in some artificial joints.
- Used to make pipes for heat exchangers (such as the large one shown here).
- Used for capacitors in electronics.

# W 74

**Tungsten** 183.8

*Swedish: "Tung sten" (heavy stone)  
Used to be called Wolframite*

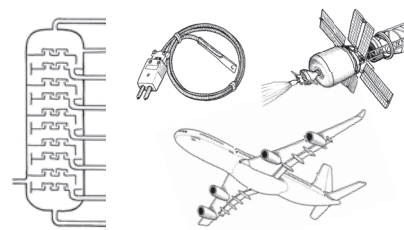


- Used for filaments in incandescent bulbs.
- Used to make drill bits and saw blades.
- Highest melting point of all the metals.
- Tungsten sulfide used as dry lubricant.
- Used as lead (Pb) substitute.

# Re 75

**Rhenium** 186.2

*Latin: "Rhenus" (Rhine River)*

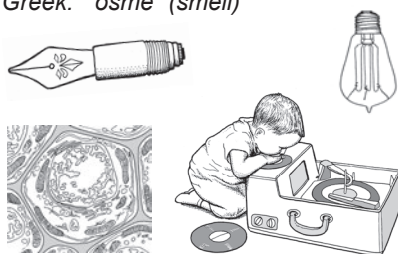


- Used in electrical switches.
- Used for thermocouples, high-temp thermometers, and oven filaments.
- Used in petroleum refining process.
- Re-W alloys used for jet engines.

# Os 76

**Osmium** 190.2

*Greek: "osme" (smell)*

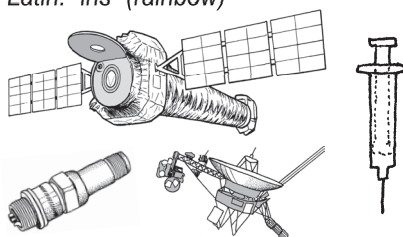


- Was used to make fountain pen points.
- Used for the first phonograph needles.
- Used for staining microscopic samples.
- Is the most dense element.
- Was used in early light bulbs.

# Ir 77

**Iridium** 192.2

*Latin: "iris" (rainbow)*

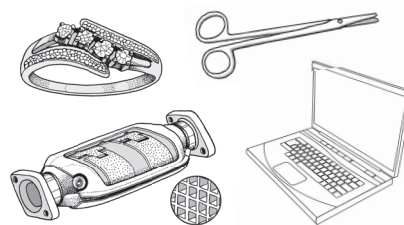


- Used to make the mirrors for Chandra X-ray observatory
- Used in airplane/helicopter spark plugs.
- Used to make hypodermic needles.
- Satellites can be powered by plutonium that is housed in iridium containers.

# Pt 78

**Platinum** 195.1

*Spanish: "platina" (silver)*

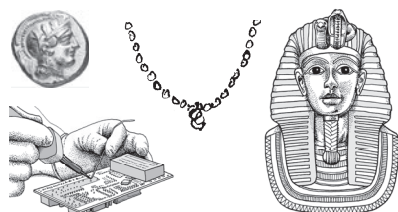


- Used in jewelry and dentistry.
- Used in computers and other devices.
- Used in catalytic converters for cars.
- Pt alloys are used to make corrosion-resistant tools.

# Au 79

**Gold** 196.9

*Old English: "gold"  
"Au" comes from Latin: "aurum"*



- Used for coins and jewelry.
- Used to make electrical circuits.
- Ancient artifacts often contain gold.
- Used as a reflective coating on the outside of large glass windows.

# Hg 80

**Mercury** 200.6

*named after the Roman god Mercury*

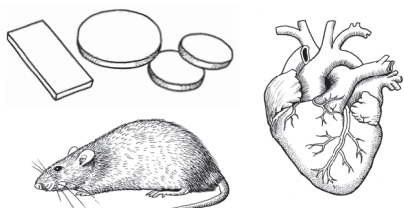


- The symbol **Hg** comes from the Latin "hydragyrum" meaning "liquid silver."
- Used in thermometers, barometers, street lights, and fluorescent bulbs.
- Was used by hat-makers in the 1800s.

# Tl 81

**Thallium** 204.4

*Greek: "thallos" (green twig)*

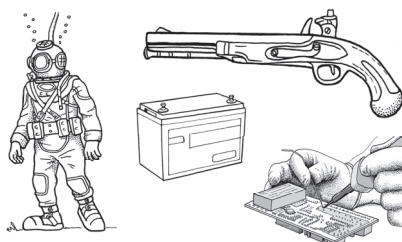


- Looks like lead and is poisonous.
- Was once used to kill pests.
- Used to diagnose heart disease.
- Used in some automatic outdoor lights.
- Used in specialty lenses.

# Pb 82

**Lead** 207.2

Ancient Anglo-Saxon: "lead"  
"Pb" comes from Latin: "Plumbum"

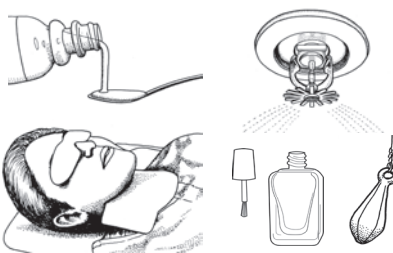


- Was used for fishing/diving weights.
- Was used to make "shot" (bullets).
- Still used in batteries for cars and boats.
- Has been used in soldering.

# Bi 83

**Bismuth** 208.9

German "weisse masse" (white mass)

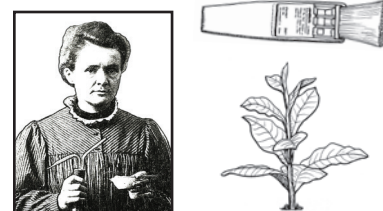


- Used in pink, liquid stomach medicines.
- Used in indoor sprinkler systems.
- Used as lead replacement for sinkers.
- Used in shields protecting from x-rays.
- Found in iridescent nail polish.

# Po 84

**Polonium** 210

named after Poland

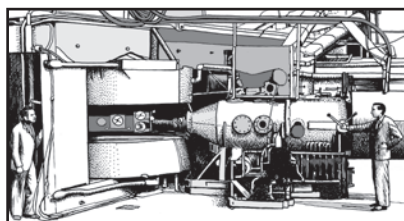


- Discovered by Marie Curie.
- Used in anti-static brushes.
- Radon from ground sticks to tobacco leaves, then decays into polonium, helping to make tobacco a carcinogen.

# At 85

**Astatine** 210

Greek: "astatos" (instable)



- Created in 1940 in the cyclotron at what is now Berkeley National Lab. They threw alpha particles (two protons and two neutrons) at bismuth atoms.
- No commercial applications.

# Rn 86

**Radon** 222

named after the element radium

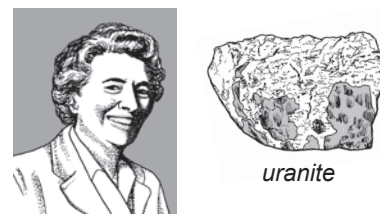


- Is the heaviest gaseous element.
- Comes up out of the ground,
- Decays into polonium.
- Probably the result of uranium decay.

# Fr 87

**Francium** 223

named after France



- Discovered in France by Marguerite Perey, a student of Marie Curie.
- Comes from the decay of U and Th (in minerals pitchblende and uranite).
- Is too unstable to be used for anything.

# Ra 88

**Radium** 226.0

Latin: "radius" (ray)

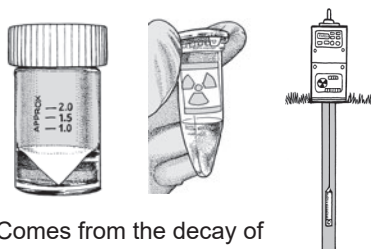


- Discovered by Marie Curie.
- Was once used to make glowing clocks.
- Used widely before it was discovered to be terribly dangerous.
- Now used to treat bone cancer because it will go to bones like calcium does.

# Ac 89

**Actinium** 227

Greek: "actinos" (ray or beam)



- Comes from the decay of uranium and thorium.
- One particular isotope is very useful in treating certain types of cancer.
- Shipped in V-shaped vials. Ac atoms collect at point of V.

# Th 90

**Thorium** 232

after the ancient Scandinavian god Thor, god of lightning and thunder



- Was used to make mantles for gas lanterns until its radioactivity was found.
- Also used to be used in welding electrodes and specialty lenses.
- One isotope is relatively stable.



**Pa**  **91**

**Protactinium** **231**

Greek: "protos" (first), plus "actinium"

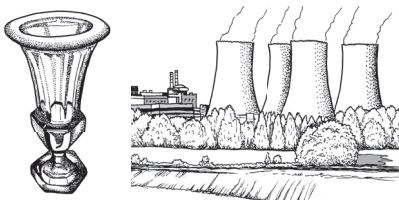
Pa  $\Rightarrow$  Ac

- Was given this name because it always decays into actinium. (Protactinium "comes first.")
- Found in nuclear waste.
- Levels of Pa and Th are studied in ocean sediments in order to learn about their history.


**U**  **92**

**Uranium** **238**

named after the planet Uranus

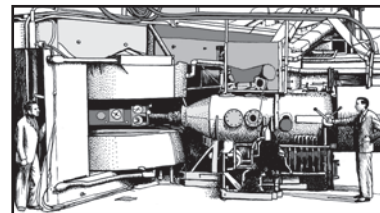


- Used as fuel in nuclear reactors.
- Depleted uranium (which is much less radioactive) is used to color glass yellow and to make metals for military vehicles.
- Primary ores: pitchblende, uranite.

**Np**  **93**

**Neptunium** **237**

named after the planet Neptune

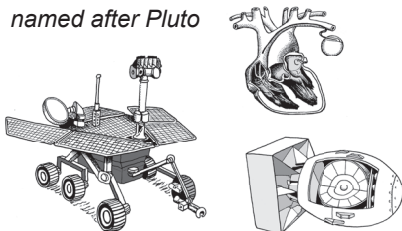


- Was manufactured in 1940 at what is now Berkeley National Lab.
- Is found in nuclear waste.
- Formation of Np from U is a step in the process of making weapons-grade Pu.

**Pu**  **94**

**Plutonium** **242**

named after Pluto

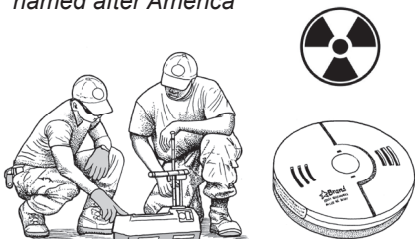


- Is made from uranium inside "breeder" nuclear reactors.
- Used in nuclear weapons.
- Was used to power the lunar modules. Now powers satellites and Mars rovers.
- Was used to power heart pacemakers.

**Am** **95**

**Americium** **243**

named after America

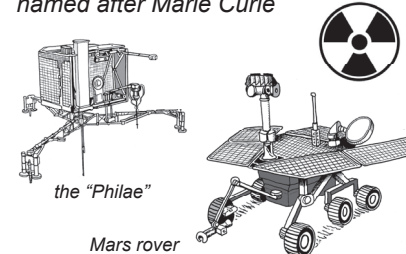


- Used in smoke detectors.
- Used in crystal research.
- Used as a source of neutrons in density gauges that search for underground water.
- Manufactured at Berkeley Lab in 1944.

**Cm** **96**

**Curium** **247**

named after Marie Curie

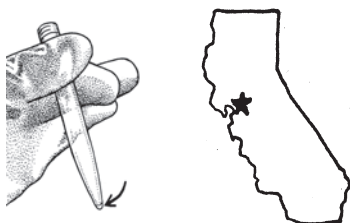


- Is used in devices that detect x-rays, and therefore can be found in the x-ray spectrometers on satellites and rovers. These devices can determine what elements are present in rocks and dirt.

**Bk**  **97**

**Berkelium** **247**

named after Berkeley, California

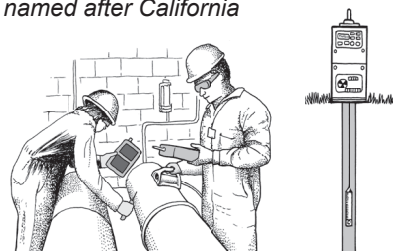


- Was made at Berkeley Lab in 1949.
- Only practical use is as a starting point for making even heavier elements.
- Like many super-heavy elements, it was discovered using a spectrometer.

**Cf**  **98**

**Californium** **251**

named after California

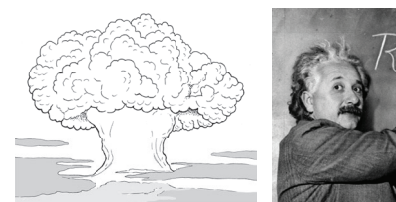


- Can be used as a portable source of neutrons in gauges that look for flaws in metal structures.
- Also used in devices that sense sources of underground water.

**Es**  **99**

**Einsteinium** **252**

named after Albert Einstein



- Discovered during the investigation of debris from the first atomic bomb.
- Einstein is famous for his equation that shows the relationship of matter to energy ( $E=mc^2$ ).

# Fm 100

**Fermium**

**257**

*named after Enrico Fermi*



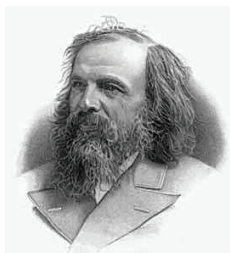
- Discovered during investigation of the debris from the first atomic bomb.
- No commercial use.
- Fermi was a physicist who studied atomic structure and radioactivity.

# Md 101

**Mendelevium**

**256**

*named after Dmitri Mendeleev*



- Made in nuclear reactors.
- No commercial use.
- Mendeleev invented the Periodic Table.

# No 102

**Nobelium**

**259**

*named after Alfred Nobel*



- Joint Institute of Nuclear Research in Dubna, Russia, given credit for discovery.
- No commercial use.
- Alfred Nobel established the Nobel Prizes using money from his invention: TNT

# Lr 103

**Lawrencium**

**262**

*named after Ernest O. Lawrence*



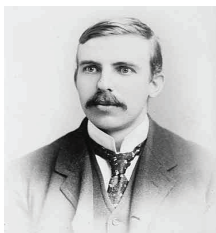
- Lawrence was the inventor of the cyclotron machine that was used to discover elements heavier than uranium.
- No commercial use.

# Rf 104

**Rutherfordium**

**261**

*named after Ernest Rutherford*



- Rutherford figured out the structure of the atom. His "gold foil" experiment showed that atoms are mostly empty space.
- No commercial use.

# Db 105

**Dubnium**

**262**

*named after Dubna, Russia*



J.I.N.R.

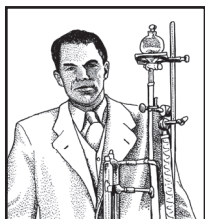
- Was made in 1968 at the Joint Institute for Nuclear Research in Dubna, Russia.
- Also made in 1978 at the Berkeley Lab.
- Russia and US share credit for discovery.
- No commercial use.

# Sg 106

**Seaborgium**

**263**

*named after Glenn T. Seaborg*



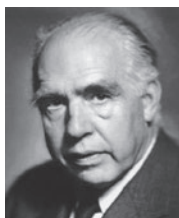
- Seaborg and his team at Berkeley Lab discovered Pu, Am, Cm, Bk, Cf, Es, Fm, Md and No.
- Is one of the few super-heavy elements that has been observed to form a molecule with other elements.

# Bh 107

**Bohrium**

**262**

*named after Niels Bohr*



- Niels Bohr discovered electron energy levels (orbital and shells).
- The longest half-life is 40 seconds.
- No commercial use.

# Hs 108

**Hassium**

**265**

*named after German state of Hesse*



- Named after the German state where the GIS Helmholtz Institute is located.
- Longest-lived isotope is 110 seconds.
- No commercial use.

# Mt 109

**Meitnerium** 268

*named after Lise Meitner*



- Meitner worked with Otto Hahn (1930-40) to discover the process of fission in uranium atoms.
- No commercial use.

# Ds 110

**Darmstadtium** 269

*named after Darmstadt, Germany*



- Discovered in 1994 at the GSI Helmholtz Institute in Darmstadt.
- No commercial use.

# Rg 111

**Roentgenium** 272

*named after Wilhelm Roentgen*



- Roentgen discovered x-rays.
- Is the heaviest member of the group (column) that contains copper, silver, gold.

# Cn 112

**Copernicium** 277

*named after Ernest O. Lawrence*



- Only element named after a scientist who was not a chemist or physicist.
- Copernicus discovered that the earth goes around the sun.
- No commercial use.

# Nh 113

**Nihonium** 284

*named after Japan (Nihon)*



- Several labs made this element but RIKEN lab in Japan given official credit.
- No commercial use.

# Fl 114

**Flervorium** 289

*named after Georgy Flyorov*



J.I.N.R.

- Georgy Flyorov was director of JINR Lab in Dubna for a number of years.
- Only 58 atoms have ever been made.
- No commercial use.

# Mc 115

**Moscovium** 288

*named after Moscow, Russia*

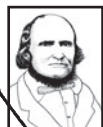
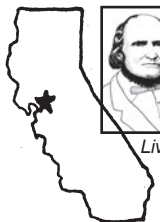


- Was first manufactured in 2004 at the JINR, which is in the state of Moscow.
- Longest-lived isotope is 1/2 second.
- No commercial use.

# Lv 116

**Livermorium** 292

*named after Livermore, CA*



Livermore



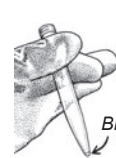
Lawrence

- Livermore, CA, got its name from its founder, Robert Livermore, a rancher.
- Lv was collaborative effort of JINR and Lawrence Livermore National Lab.
- No commercial use.

# Ts 117

**Tennessine** 294

*named after state of Tennessee*



- Manufacturing of Ts was a collaboration between JINR (Russia) and Lawrence Livermore National Lab (US).
- Made from Bk atoms that were made at Oak Ridge National Lab in Tennessee.
- No commercial use.