CK

A FAST-PACED CARD GAME ABOUT THE ELEMENTS

<u>You will need</u>: Scissors, photocopies of the pattern pages printed onto card stock

<u>TIP</u>: If you will be making mulitple copies for a large classroom, consider making each set a different color. If you can't do this, then be careful to mark each set somehow, even if it is just a colored dot on the back. You will be glad you did this when the sets get all mixed up by the end of the class.



NOTE: Unfortunately, 118 divided by 9 leaves remainder 1. That means 1 card was doomed to be on the last page by itself. It is printed here on the instructions page, leaving the decision in your hands whether to include it in the game or not. If you'd like to use it, just copy this page onto card stock along with the other pattern pages.

Set up:

Cut apart the cards, then shuffle them thoroughly.

How to play:

1) The object of the game is to be the first player to collect six cards.

2) Decide which player will be the "caller." This player must read from the list below instead of being one of the card players. If an adult is supervising the game, this is the obvious adult job. An adult caller may want to choose particular attributes from the list below to emphasize facts recently learned. It is easiest to go down the list in order, but the caller need not go in order, and may also use items from the list more than once. Feel free to add your own ideas and write additional clues.

3) Each card player receives five cards, which he places face up in front of him. The rest of the cards go face down in a draw pile. The caller reads the first clue. Each player looks at their five cards to see if they have a card that matches the clue. If they do, they slap their hand down on the card. The caller looks to see who is the first player to slap their hand down. That player then shows the card under their hand. If the caller agrees that this card qualifies, then the player may remove that card from their line-up and put it face down into a "keeper" pile. Then they draw a card from the draw pile to replace that card, giving them five face-up cards once again.

4) The caller then reads off another clue from the list and the game continues in this manner until one player has six cards in their "keeper" pile.

- 5) If no player has a card that qualifies for a certain clue, the caller simply goes on to the next clue.
- 6) If you reach the end of the list, just start over at the beginning.
- 7) One round takes 5-20 minutes to play. You can switch callers between rounds if you want to.

NOTE: Some of these clues require the students to look at the atomic weight, or "mass," of the element. (Weight and mass are not really the same thing, but in this case the words can be used interchangeably. Kids seem to prefer "weight" to "mass.") The atomic mass is listed in smaller print right under the atomic number. It is basically the number of protons and neutrons added together. Electrons are so small they add almost nothing to the total mass. The reason that some of the masses are decimal numbers is that scientists measured many atoms, then took a mathematical average. Since a small percentage of atoms have one or two more (or less) neutrons, the average comes out to a decimal number. For example, if you weigh ten atoms of neon and get these results: 20, 20, 20, 20, 20, 20, 20, 21, 21. The average is 20.2, so this is the atomic mass listed for neon. Most neon atoms have 10 protons and 10 neutrons, but once in a while you will meet a neon atom with 10 protons 11 neutrons.

QUICK SIX CLUES

The clues are in groups of ten just to make them easier to read (so you don't lose your place so easily).

Number has a 3 in it Name has two syllables Used in lasers Has something to do with the color green Named after Scandinavia, or a place in Scandinavia Has something to do with teeth Starts with the letter C Number has a 5 in it Name has something to do with color Used to make tools of some kind

Is named after a city (not a country) Name has three syllables Is used to make jewelry Named after a country Used for something that burns Named after something in the solar system Number has a 7 in it Is named after a country (not a city) Used in fireworks Has something to do with bones

Name starts with a vowel Is found in gemstones Used in steel production Used to repair the human body Used in light bulbs Is found as a gas in the air around us Has something to do with eyes Conducts electricity Last three letters of the name are I-U-M Name is from a Latin word

Is used in batteries or fuel Has something to do with glass First letter of name does not match first letter of the symbol Is found in some kind of gemstone Name begins with the letter S Name comes from a chemical compound Name starts with the "K" sound (C or K) Is used in magnets of any kind Used in something that makes light Has been used to make coins

Contains one of these letters: X, Y, or Z Name has four syllables Number has a 1 in it Does not bond with any other element Found in some kind of First Aid product (antiseptics, bandages, etc) Quick Six clues page 2

Name comes from a mineral or chemical compound Has an atomic number less than that of tin Has the word "light" or "lights" in the description Name ends with –ine Has an atomic number between 50 and 60

Atomic number has a 3 in it Name has two syllables Used in hygeine product Has something to do with the color green Named after someplace in Scandinavia Has something to do with teeth Named after a Greek god or goddess Is a transition metal Starts with the letter C Is in the same row as gold on the Periodic Table

Used in some kind of engine Atomic number has a 5 in it Used to make tools of some kind Is named after a city (not a country) Is an alkali earth metal Is radioactive Name has three syllables Is used to make jewelry Used for something that burns Is a non-metal

Atomic mass is less than 30 Named after something in the solar system Atomic number has a 7 in it Is on the edge of the Periodic Table Atomic mass is between 50 and 70 Named after Ytterby, Sweden Is a true metal or semi-metal Is named after a country (not a city) Used in fireworks Atomic number has three digits

Can be extracted from monazite sands Is in the actinide series Has something to do with bones Name starts with a vowel Is in the same row as molybdenum on the Periodic Table Gemstones are made from it Named after a famous scientist Has an atomic number greater than that of tungsten Used to color glass Name has four syllables Quick Six clues page 3

Atomic number has a 0 in it Used in steel production or in bronze or copper Used to repair the human body in some way Is in the same column as helium on the Periodic Table Used in light bulbs Atomic mass is greater than 100 Is found as a gas in the air around us Has something to do with eyes Atomic number has a 9 in it Is in the lanthanide series

Conducts electricity Last three letters of the name are I U M Is in the same row as iron on the Periodic Table Has no commercial use Is made in nuclear reactors Name is from a Latin word First letter of name does not match first letter of the symbol The atomic mass listed on the card is exactly double the atomic number Name comes from a Greek word or words Used in anything with a screen-- televsions, computuers, phones

Is in the third row of the periodic table Name starts with the letter *u* Used for coins Unstable; only exists for a short time Used in catalytic converters Has a *y* in its name Is in the third column of the periodic table Used in lights Made in nuclear reactors Name comes from a German word or words

All the digits of the atomic number are the same Used in batteries Name has something to do with a color The digits of its atomic number add up to 10 Used in magnets Has an x in its name Atomic number is a prime number Name starts with the letter mExists for only seconds or minutes (elements beyond the actinide series) Has more than 4 vowels in its name (y is a vowel)

Is in the first column of the periodic table Name ends in "–on" Name starts with the letter *s* Has an atomic mass less than 10 Name has less than 5 letters The sum of the atomic number and atomic mass is between 100 and 200



- H₂O is water. • H₂O₂ is hydrogen peroxide.
- Found in gun powder.

4.0 Lithium 6.9 Greek: "lithos" (stone) A WHITE GREASE · Used in batteries, lubricants, medicines, red fireworks, and nuclear bombs. · Is never found by itself in nature, but is always in a compound. 10.8 Carbon 12.0 Latin: "carbo" (charcoal) • Diamonds, graphite (pencil "lead") and coal are all made of carbon. Carbon makes long chains (polymers) that are the basis of fossil fuels and plastics. Carbon is necessary for organic molecules found in living organisms. 15.9 18.9 Fluorine Latin: "fluere" (to flow) · Combines with Ca to make fluorite. · Is put into toothpaste to fight cavities. Combines with C to make Teflon.[®] · Used as flux in steel making (makes hot metal flow better).



"Quick Six" pattern page 2

Copy onto white card stock



44.9

51.9

Ballenti







88 9









138.9

144.2



168.9





3.7.



it will go to bones like calcium does.

- Polonium German" "weisse masse" (white mass) named after Poland Discovered by Marie Curie. • Used in pink, liquid stomach medicines. · Used in anti-static brushes. • Used in indoor sprinkler systems. Radon from ground sticks to tobacco · Used as lead replacement for sinkers. leaves, then decays into polonium, · Used in shields protecting from x-rays. helping to make tobacco a carcinogen. · Found in iridescent nail polish. Kn 😒 222 Francium named after the element radium named after France ۱D Π $\Box \Box$ 30. uranite 13 · Discovered in France by Marguerite · Is the heaviest gaseous element. Perey, a student of Marie Curie. · Comes up out of the ground, • Comes from the decay of U and Th · Decays into polonium. (in minerals pitchblende and uranite). • Probably the result of uranium decay. · Is too unstable to be used for anything. I h 😒 9 Actinium 227 Thorium Greek: "actinos" (ray or beam) after the ancient Scandinavian god Thor. god of lightning and thunder
- · Comes from the decay of uranium and thorium.

Radon

- · One particular isotope is very useful
- in treating certain types of cancer.
- Shipped in V-shaped vials. Actinium
- atoms collect at point of V.

Copy onto white card stock

• Was used for mantles for gas lanterns

· Was also used in welding electrodes

until found to be mildly radioactive.

· One isotope is relatively stable.

and specialty lenses.

210

223

232

"Quick Six" pattern page 10



Protactinium 231 Greek: "protos" (first), plus "actinium"



- · Was given this name because it always decays into actinium. (Protactinium "comes first.")
- · Found in nuclear waste.

Plutonium

named after Pluto

· Levels of Pa and Th are studied in ocean sediments in order to learn about the history of the sediments.

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247





238

Neptunium

named after the planet Neptune

Uranium

named after the planet Uranus



- · Used as fuel in nuclear reactors.
- Depleted uranium (which is much less
- radioactive) is used to color glass yellow
- and to make metals for military vehicles.
- Primary ores: pitchblende, uranite.



- Used as a source of neutrons in density
- gauges that search for underground water. Manufactured at Berkeley Lab in 1944.



Californium named after California



· Can be used as a portable source of neutrons in gauges that look for flaws in metal structures.

· Also used in devices that sense sources of underground water.



· Is used in devices that detect x-rays, and therefore can be found in the x-ray spectrometers on satellites and rovers. These devices can determine what elements are present in rocks and dirt.





252

237

Einsteinium named after Albert Einstein



· Discovered during the investigation of debris from the first atomic bomb. Einstein is famous for his equation that shows the relationship of matter to energy (E=mc²).

 Is made from uranium inside "breeder" nuclear reactors.

- · Used in nuclear weapons.
- · Was used to power the lunar modules.
- Now powers satellites and Mars rovers.
- · Was used to power heart pacemakers.



Berkelium named after Berkeley, California



 Was made at Berkeley Lab in 1949. · Only practical use is as a starting point for making even heavier elements. · Like many super-heavy elements, it was discovered using a spectrometer.





form a molecule with other elements.



No commercial use.



Roentgenium

named after Wilhelm Roentgen



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272

named after Georgy Flyorov





Joint Institute of Nuclear Research

- Geory Flyorov was director of JINR Lab in Dubna for a number of years. • Only 58 atoms have ever been made.
- · No commercial use.



294 Tennessine named after state of Tennessee



- Manufacturing of Ts was a collaboration between JINR (Russia) and Lawrence Livermore National Lab (US).
- Made from Bk atoms that were made at
- Oak Ridge National Lab in Tennessee. • No commercial use.

JINR, which is in the state of Moscow.

• Longest-lived isotope is 1/2 second.

• No commercial use.



 Several labs made this element but RIKEN lab in Japan was given official



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· Livermore, CA, got its name from its founder, Robert Livermore, a rancher. Ly was a collaborative effort of JINR and Lawrence Livermore National Lab.





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